

## DESCRIPTION

BENZIMIDAZOLE DERIVATIVE

## 5           Technical Field

The present invention relates to a benzimidazole derivative, and more particularly, to a benzimidazole derivative useful as an inhibitor of human chymase activity.

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## Background Art

Chymase is a neutral protease present in mast cell granules, and is intimately involved in various biological reactions participated in by mast cells. For example, chymase has been reported to have various actions, including the promotion of degranulation from mast cells, activation of Interleukin- $1\beta$  (IL- $1\beta$ ), activation of matrix protease, decomposition of fibronectin and type IV collagen, promotion of the liberation of transforming growth factor- $\beta$  (TGF- $\beta$ ), activation of substance P and vasoactive intestinal polypeptide (VIP), conversion from angiotensin I (Ang I) to angiotensin II (Ang II), and conversion of endothelin.

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On the basis of the above, inhibitors of said chymase activity are considered to be promising as preventive and/or therapeutic agents against respiratory diseases such as bronchial asthma, inflammatory and allergic diseases such as allergic rhinitis, atopic dermatitis and urticaria, cardiovascular diseases such as sclerosing vascular lesions, vasoconstriction, peripheral circulatory disorders, renal insufficiency and cardiac insufficiency, and bone and cartilage metabolic diseases such as rheumatoid arthritis and osteoarthritis.

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Although known examples of chymase activity inhibitors of the prior art include a triazine derivative (Japanese Unexamined Patent Publication No. 8-208654),

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hydantoin derivative (Japanese Unexamined Patent Publication No. 9-31061), imidazolidine derivative (International Publication No. WO96/04248), quinazoline derivative (International Publication No. WO97/11941),  
5 heterocyclic amide derivative (International Patent Publication No. WO96/33974), cefam compound (Japanese Unexamined Patent Publication No. 10-087493), phenol derivative (Japanese Unexamined Publication No. 10-087567), heterocyclic amide compound (International  
10 Publication No. WO98/18794), acetoamide derivative (International Publication No. WO98/09949), heterocyclic amide compound (Japanese Unexamined Publication No. 10-007661), acid anhydride derivative (Japanese Unexamined Patent Publication No. 11-049739), heterocyclic amide  
15 compound (International Publication No. WO99/32459) and acetoamide derivative (International Publication No. WO99/41277), these compounds and the compound of the present invention are completely different structurally.

20 The chymase inhibitor compounds disclosed thus far have lacked usefulness as a result of having inadequate activity or being structurally unstable. However, the compound of the present invention has extremely high activity and demonstrates superior kinetics in the blood, making it highly useful as a drug.

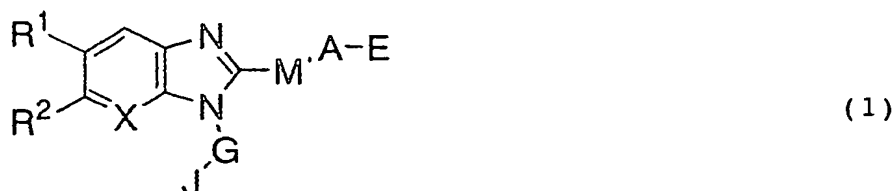
25 On the other hand, an example of a technology related to the compound of the present invention is described in the specification of US Patent No. 5124336. A benzimidazole derivative is described in said  
30 specification as a compound that has thromboxane receptor antagonistic activity. However, the compound described in said specification is not disclosed as having a heteroaryl group substituted in the benzimidazole skeleton, and there is also no description of human  
35 chymase activity of said compound. In addition, although a benzimidazole compound is also described as an antitumor agent in Japanese Unexamined Patent Publication No. 01-265089, there is no mention of human chymase

inhibitory activity.

#### Disclosure of the Invention

The object of the present invention is to provide a novel compound capable of being a human chymase activity inhibitor that can be applied clinically.

As a result of repeated and earnest research to achieve the above object, the inventors of the present invention found a benzimidazole derivative or its medically acceptable salt, represented by the following formula (1), that has a structure that is completely different from known compounds, thereby leading to completion of the present invention:



wherein,  $R^1$  and  $R^2$  may be the same or different and each independently represents a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, a hydroxyl group, an alkyl group having 1-4 carbon atoms, an alkoxy group having 1-4 carbon atoms, or  $R^1$  and  $R^2$  together represent  $-O-CH_2-O-$ ,  $-O-CH_2CH_2-O-$  or  $-CH_2CH_2CH_2-$  (these groups may be substituted by one or more alkyl groups having 1-4 carbon atoms);

A represents a substituted or unsubstituted, linear, cyclic or branched alkylene or alkenylene group having 1-7 carbon atoms which may be interrupted by one or more of  $-O-$ ,  $-S-$ ,  $-SO_2-$  and  $-NR^3-$  (where  $R^3$  represents a hydrogen atom or linear or branched alkyl group having 1-6 carbon atoms); the substituent that can be possessed by these groups is selected from a halogen atom, hydroxyl group, nitro group, cyano group, linear or branched alkyl group having 1-6 carbon atoms, linear or branched alkoxy group having 1-6 carbon atoms (including the case in which two adjacent groups form an acetal bond, namely including the

case in which the alkyl portions of geminal two alkoxy groups are connected to form a ring), a linear or branched alkylthio group having 1-6 carbon atoms, a linear or branched alkylsulfonyl group having 1-6 carbon atoms, a linear or branched acyl group having 1-6 carbon atoms, a linear or branched acylamino group having 1-6 carbon atoms, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, and a phenoxy group that may be substituted by one or more halogen atoms; and, one or more of these substituents may each independently be bonded to optional positions of the alkylene or alkenylene group;

E represents a  $-\text{COOR}^3$ ,  $-\text{SO}_3\text{R}^3$ ,  $-\text{CONHR}^3$ ,  $-\text{SO}_2\text{NHR}^3$ , tetrazole-5-yl group, a 5-oxo-1,2,4-oxadiazole-3-yl group or a 5-oxo-1,2,4-thiadiazole-3-yl group (where  $\text{R}^3$  is as defined above);

G represents a substituted or unsubstituted, linear or branched alkylene group having 1-6 carbon atoms which may be interrupted by one or more of  $-\text{O}-$ ,  $-\text{S}-$ ,  $-\text{SO}_2-$  and  $-\text{NR}^3-$  (where,  $\text{R}^3$  is as defined above. Where these atoms or atomic groups exist, they are not bonded directly to the benzimidazole ring.); and, the substituent that can be possessed by said alkylene group is selected from a halogen atom, a hydroxyl group, a nitro group, a cyano group, a linear or branched alkyl group having 1-6 carbon atoms, a linear or branched alkoxy group having 1-6 carbon atoms (including the case in which two adjacent groups form an acetal bond), a trihalomethyl group, a trihalomethoxy group, a phenyl group, and an oxo group;

M represents a single bond or  $-\text{S}(\text{O})_m-$ , where m is an integer of 0-2;

J represents a substituted or unsubstituted heterocyclic group having 4-10 carbon atoms and containing one or more hetero atoms selected from the group consisting of an oxygen atom, a nitrogen atom and a sulfur atom on its ring, with the proviso that an imidazole ring is excluded; the substituent that can be

possessed by said aromatic heterocyclic group is selected from a halogen atom, a hydroxyl group, a nitro group, a cyano group, a linear or branched alkyl group having 1-6 carbon atoms, a linear or branched alkoxy group having 1-6 carbon atoms (including the case in which two adjacent groups form an acetal bond), a linear or branched alkylthio group having 1-6 carbon atoms, a linear or branched alkylsulfonyl group having 1-6 carbon atoms, a linear or branched acyl group having 1-6 carbon atoms, a linear or branched acylamino group having 1-6 carbon atoms, a substituted or unsubstituted anilide group, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, a COOR<sup>3</sup> group, and a phenoxy group that may be substituted by one or more halogen atoms; and, one or more of these substituents may be substituted at optional positions on the ring; and, X represents a methine group (-CH=) or nitrogen atom.

## Best Mode for Carrying Out the Invention

The substituents in the compound of the present invention represented by the above formula (1) are as indicated below.

R<sup>1</sup> and R<sup>2</sup> may be the same or different and each independently represents a hydrogen atom, a halogen atom, trihalomethyl group, a cyano group, a hydroxyl group, an alkyl group having 1-4 carbon atoms, an alkoxy group having 1-4 carbon atoms. Alternatively, R<sup>1</sup> and R<sup>2</sup> together represent -O-CH<sub>2</sub>-O-, -O-CH<sub>2</sub>CH<sub>2</sub>-O- or -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, and in this case, these groups may be substituted by one or more alkyl groups having 1-4 carbon atoms.

Specific examples of the alkyl groups having 1-4 carbon atoms as R<sup>1</sup> and R<sup>2</sup> include a methyl group, an ethyl group, an n- or i-propyl group and an n-, i-, s- or t-butyl group. A preferable example is a methyl group. Specific examples of alkoxy groups having 1-4 carbon atoms include a methoxy group, an ethoxy group, an n- or

i-propoxy group and an n-, i-, s- or t-butoxy group.

Preferable examples of  $R^1$  and  $R^2$  include a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, a hydroxyl group, an alkyl group having 1-4 carbon atoms and an alkoxy group having 1-4 carbon atoms. More preferable examples include a hydrogen atom, a halogen atom, a trihalomethyl group, a cyano group, an alkyl group having 1-4 carbon atoms and an alkoxy group having 1-4 carbon atoms, still more preferable examples include a hydrogen atom, chlorine atom, a fluorine atom, a trifluoromethyl group, methyl group, a methoxy group and an ethoxy group, while particularly preferable examples include a hydrogen atom, a methyl group and a methoxy group.

A represents a substituted or unsubstituted, linear, cyclic or branched alkylene or alkenylene group having 1-7 carbon atoms. Examples of the unsubstituted, linear, cyclic or branched alkylene group having 1-7 carbon atoms include a methylene group, an ethylene group, an n- or i-propylene group, a 2,2-dimethylpropylene group, an n-, i- or t-butylene group, a 1,1-dimethylbutylene group, an n-pentylene group and a cyclohexylene group. More preferable examples include an ethylene group, an n-propylene group, a 2,2-dimethylpropylene group and an n- or t-butylene group. Still more preferable examples include an n-propylene group and a 2,2-dimethylpropylene group. A particularly preferable example is an n-propylene group. Examples of the unsubstituted linear or branched alkenylene group having 1-7 carbon atoms include a vinylene group, a propenylene group, a butenylene group and a pentenylene group.

Although said alkylene group or alkenylene group may be interrupted by one or more of -O-, -S-, -SO<sub>2</sub>- and -NR<sup>3</sup>- (where R<sup>3</sup> represents a hydrogen atom or linear or branched alkyl group having 1-6 carbon atoms), these atoms or atomic groups are not bonded directly to M. Specific examples include interrupted ethylene groups, n-

propylene groups or n- or t-butylene groups. More specific examples include  $-\text{CH}_2\text{OCH}_2-$ ,  $-\text{CH}_2\text{OCH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{SCH}_2-$ ,  $-\text{CH}_2\text{SCH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{SO}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{SO}_2\text{CH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{NR}^4\text{CH}_2-$  and  $-\text{CH}_2\text{NR}^4\text{CH}_2\text{CH}_2-$ . Preferable examples  
5 include  $-\text{CH}_2\text{OCH}_2-$ ,  $-\text{CH}_2\text{SCH}_2-$  and  $-\text{CH}_2\text{SO}_2\text{CH}_2-$ .

The substituent groups that can be possessed by said alkylene group is selected from a halogen atom, a hydroxyl group, a nitro group, a cyano group, a linear or branched alkyl group having 1-6 carbon atoms, a linear or  
10 branched alkoxy group having 1-6 carbon atoms (including the case in which two adjacent groups form an acetal bond), a linear or branched alkylthio group having 1-6 carbon atoms, a linear or branched alkylsulfonyl group having 1-6 carbon atoms, linear or branched acyl group  
15 having 1-6 carbon atoms, a linear or branched acylamino group having 1-6 carbon atoms, a trihalomethyl group, a trihalomethoxy group, a phenyl group, an oxo group, and a phenoxy group that may be substituted by one or more halogen atoms. One or more of these substituents may  
20 each be independently bonded to optional positions of the alkylene group or alkenylene group.

Examples of the halogen atom include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom. Preferable examples are a fluorine atom and a chlorine  
25 atom.

Specific examples of the linear or branched alkyl group having 1-6 carbon atoms include a methyl group, an ethyl group, an n- or i-propyl group and an n-, i-, s- or t-butyl group, while preferable examples are a methyl  
30 group and an ethyl group. A more preferable example is a methyl group.

Specific examples of the linear or branched alkoxy group having 1-6 carbon atoms include a methoxy group, an ethoxy group, an n- or i-propoxy group and an n-, i-, s-  
35 or t-butoxy group, while preferable examples are a methoxy group and an ethoxy group. A more preferable example is a methoxy group.

Specific examples of the linear or branched alkylthio group having 1-6 carbon atoms include a methylthio group, an ethylthio group, an n- or i-propylthio group, and an n-, i-, s- or t-butylthio group, and preferable examples are a methylthio group and an ethylthio group. A more preferable example is a methylthio group.

Specific examples of the linear or branched alkylsulfonyl group having 1-6 carbon atoms include a methylsulfonyl group, an ethylsulfonyl group, an n- or i-propylsulfonyl group and an n-, i-, s- or t-butylsulfonyl group, and preferable examples are a methylsulfonyl group and an ethylsulfonyl group. A more preferable example is a methylsulfonyl group.

Examples of the linear or branched acyl group having 1-6 carbon atoms include an acetyl group, an ethylcarbonyl group, an n- or i-propylcarbonyl group and an n-, i-, s- or t-butylcarbonyl group, and preferable examples are an acetyl group and an ethylcarbonyl group. A more preferable example is an acetyl group.

Specific examples of the linear or branched acylamino group having 1-6 carbon atoms include an acetylamino group, an ethylcarbonylamino group, an n- or i-propylcarbonylamino group and an n-, i-, s- or t-butylcarbonylamino group, and preferable examples are an acetylamino group and an ethylcarbonylamino group. A more preferable example is an acetylamino group.

Specific examples of the trihalomethyl group are a trifluoromethyl group, a tribromomethyl group and a trichloromethyl group. A preferable example is a trifluoromethyl group.

In particular, A is preferably a substituted or unsubstituted, linear, cyclic or branched alkylene group having 1-7 carbon atoms (although it may be interrupted by one or more of -O-, -S-, -SO<sub>2</sub>- and -NR<sup>3</sup>- (where NR<sup>3</sup> is as defined above), these atoms or atomic groups not being bonded directly to M). Preferable examples include



-CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>C(=O)CH<sub>2</sub>-, -CH<sub>2</sub>OCH<sub>2</sub>-, -CH<sub>2</sub>SCH<sub>2</sub>-,  
 -CH<sub>2</sub>S(=O)CH<sub>2</sub>-, -CH<sub>2</sub>CF<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>SO<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-,  
 -CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>SO<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>C(=O)CH<sub>2</sub>CH<sub>2</sub>-,  
 -CH<sub>2</sub>C(=O)(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>-, and -CH<sub>2</sub>C(=O)C(=O)CH<sub>2</sub>-. More  
 5 preferable examples are -CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-,  
 -CH<sub>2</sub>C(=O)CH<sub>2</sub>-, -CH<sub>2</sub>OCH<sub>2</sub>-, -CH<sub>2</sub>SCH<sub>2</sub>-, -CH<sub>2</sub>S(=O)CH<sub>2</sub>-,  
 -CH<sub>2</sub>CF<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>SO<sub>2</sub>CH<sub>2</sub>- and -CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>-. Still more  
 preferable examples are -CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>- and  
 -CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>-. A particularly preferable example is  
 10 -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-.

E represents a -COOR<sup>3</sup>, -SO<sub>3</sub>R<sup>3</sup>, -CONHR<sup>3</sup>, -SO<sub>2</sub>NHR<sup>3</sup>,  
 tetrazole-5-yl, 5-oxo-1,2,4-oxadiazole-3-yl or 5-oxo-  
 1,2,4-thiadiazole-3-yl group (where, R<sup>3</sup> represents a  
 hydrogen atom or linear or branched alkyl group having 1-  
 15 6 carbon atoms).

Examples of R<sup>3</sup> include a hydrogen atom, a methyl  
 group, an ethyl group, an n- or i-propyl group and an n-,  
 i-, s- or t-butyl group. Preferable examples are a  
 hydrogen atom, a methyl group and an ethyl group. A  
 20 particularly preferable example is a hydrogen atom.

In particular, preferable examples of E are -COOR<sup>3</sup>,  
 -SO<sub>3</sub>R<sup>3</sup>, and tetrazole-5-yl groups. A more preferable  
 example is a -COOR<sup>3</sup> group. A particularly preferable  
 example is a -COOH group.

G represents a substituted or unsubstituted, linear  
 or branched alkylene group having 1-6 carbon atoms which  
 may be interrupted by one or more of -O-, -S-, -SO<sub>2</sub>- and  
 -NR<sup>3</sup>-. Here, R<sup>3</sup> is as defined above. In addition, in  
 the case of containing these hetero atoms or atomic  
 30 groups, they are not directly bonded to the benzimidazole  
 ring. The substituent that can be possessed by the  
 alkylene group is selected from a halogen atom, a  
 hydroxyl group, a nitro group, a cyano group, a linear or  
 branched alkyl group having 1-6 carbon atoms, a linear or  
 35 branched alkoxy group having 1-6 carbon atoms (including  
 the case in which two adjacent groups form an acetal  
 bond), a trihalomethyl group, a trihalomethoxy group, a

phenyl group and an oxo group. Specific examples of G include  $-\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{CO}-$ ,  $-\text{CH}_2\text{CH}_2\text{O}-$ ,  $-\text{CH}_2\text{CONH}-$ ,  $-\text{CO}-$ ,  $-\text{SO}_2-$ ,  $-\text{CH}_2\text{SO}_2-$ ,  $-\text{CH}_2\text{S}-$  and  $-\text{CH}_2\text{CH}_2\text{S}-$ , while preferable examples are  $-\text{CH}_2-$ ,  $-\text{CH}_2\text{CH}_2-$ ,  $-\text{CH}_2\text{CO}-$  and  $-\text{CH}_2\text{CH}_2\text{O}-$ . More preferable examples are  $-\text{CH}_2-$  and  $-\text{CH}_2\text{CH}_2-$ , and a particularly preferable example is  $-\text{CH}_2-$ . These groups are bonded on the left hand side to position 1 (N atom) of the benzimidazole ring, while on the right hand side to J.

M represents a single bond or  $-\text{S}(\text{O})_m-$ , where m represents an integer of 0-2. Preferable examples of M are  $-\text{S}-$  and  $-\text{SO}_2-$ . A particularly preferable example is  $-\text{S}-$ .

J represents a substituted or unsubstituted heterocyclic group having 4-10 carbon atoms and containing one or more hetero atoms selected from the group consisting of an oxygen atom, a nitrogen atom and a sulfur atom on its ring. However, an imidazole ring is excluded. In addition, J is limited to that which can be chemically synthesized.

Specific examples of the unsubstituted heterocyclic groups having 4-10 carbon atoms and containing one or more hetero atoms on its ring selected from the group consisting of an oxygen atom, a nitrogen atom and a sulfur atom include a pyridyl group, a furyl group, a thienyl group, a thiazolyl group, a pyrimidinyl group, an oxazolyl group, an isooxazolyl group, a benzofuryl group, a benzimidazolyl group, a quinolyl group, an isoquinolyl group, a quinoxalinyl group, a benzoxadiazolyl group, a benzothiaziazolyl group, an indolyl group, a benzothiazolyl group, a benzothienyl group and a benzoisooxazolyl group. A preferable example is a bicyclic heterocyclic ring. More preferable examples are a benzofuryl group, a benzoimidazolyl group, a quinolyl group, an isoquinolyl group, a quinoxalinyl group, a benzoxadiazolyl group, a benzothiazolyl group, an indolyl group, a benzothiazolyl group, a benzothienyl group and a

benzoisooxazolyl group, while a particularly preferable example is a benzothienyl group or an indolyl group.

5       The substituent groups that can be possessed by the aromatic heterocyclic group is selected from a halogen atom, a hydroxyl group, a nitro group, a cyano group, a linear or branched alkyl group having 1-6 carbon atoms, a linear or branched alkoxy group having 1-6 carbon atoms (including the case in which two adjacent groups form an acetal bond), a linear or branched alkylthio group having 1-6 carbon atoms, a linear or branched alkylsulfonyl group having 1-6 carbon atoms, a linear or branched acyl group having 1-6 carbon atoms, a linear or branched acylamino group having 1-6 carbon atoms, a substituted or unsubstituted anilide group, a trihalomethyl group, a trihalomethoxy group, a phenyl group, and a phenoxy group that may be substituted by one or more halogen atoms. One or more of these substituents groups may each independently be bonded to optional positions of the ring.

20       Examples of the halogen atom are a fluorine atom, a chlorine atom, a bromine atom and an iodine atom. Preferable examples are a fluorine atom and a chlorine atom.

25       Specific examples of the linear or branched alkyl groups having 1-6 carbon atoms include a methyl group, an ethyl group, an n- or i-propyl group and an n-, i-, s- or t-butyl group, and preferable examples are a methyl group and an ethyl group. A more preferable example is a methyl group.

30       Specific examples of the linear or branched alkoxy groups having 1-6 carbon atoms include a methoxy group, an ethoxy group, an n- or i-propyloxy group, an n-, i-, s- or t-butyloxy group and a methylenedioxy group, and preferable examples are a methoxy group and an ethoxy group. A more preferable example is a methoxy group.

35       Specific examples of the linear or branched alkylthio group having 1-6 carbon atoms include a

methylthio group, an ethylthio group, an n- or i-propylthio group and an n-, i-, s- or t-butylthio group, and preferable examples are a methylthio group and an ethylthio group. A more preferable example is a methylthio group.

Specific examples of the linear or branched alkylsulfonyl group having 1-6 carbon atoms include a methylsulfonyl group, an ethylsulfonyl group, an n- or i-propylsulfonyl group and an n-, i-, s- or t-butylsulfonyl group, and preferable examples are a methylsulfonyl group and an ethylsulfonyl group. A more preferable example is a methylsulfonyl group.

Specific examples of the linear or branched acyl group having 1-6 carbon atoms include an acetyl group, an ethylcarbonyl group, an n- or i-propylcarbonyl group and an n-, i-, s- or t-butylcarbonyl group, and preferable examples are an acetyl group and an ethylcarbonyl group. A more preferable example is an acetyl group.

Specific examples of the linear or branched acylamino group having 1-6 carbon atoms include an acetylamino group, an ethylcarbonylamino group, an n- or i-propylcarbonylamino group and an n-, i-, s- or t-butylcarbonylamino group, and preferable examples are an acetylamino group and an ethylcarbonylamino group. A more preferable example is an acetylamino group.

Specific examples of the trihalomethyl group include a trifluoromethyl group, a tribromomethyl group and a trichloromethyl group.

X represents a  $-CH=$  group or nitrogen atom, and a preferable example is a  $-CH=$  group.

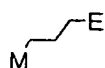
Preferable examples of the compounds represented by the above formula (1) include various groups of compounds composed by combining each of the groups previously described as preferable examples. Although there is no intention of limiting these groups, those described in the following table are particularly preferable. In particular, preferable examples of those compounds in the

table include compound Nos. 34, 38, 39, 41, 42, 52, 54, 56, 58, 59, 63, 135, 137, 148, 152, 154, 244, 340, 436, 514, 519, 521, 532, 534, 536, 538, 615, 628, 1112 and 1114.

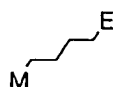
5 Furthermore, A1 through A3 and J1 through J32 in the following table are groups represented with the following formulas. In the formulas, although E, G, M, m and X are as defined above, they are described hereinbelow using representative examples, namely E is COOH, G is CH<sub>2</sub>, M is S (m being 0) or a single bond (indicated with "-" in the  
10 table) and X is -CH=. However, it is not intended that the present invention is limited to these compounds.



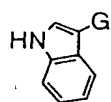
A1



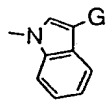
A2



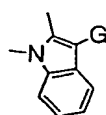
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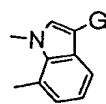
J1



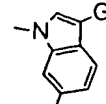
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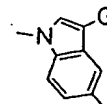
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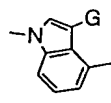
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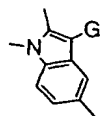
J5



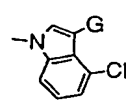
J6



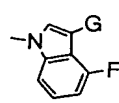
J7



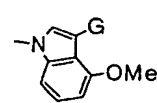
J8



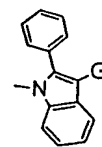
J9



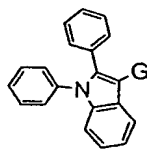
J10



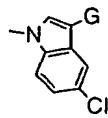
J11



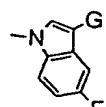
J12



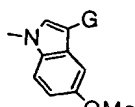
J13



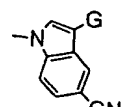
J14



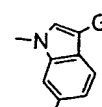
J15



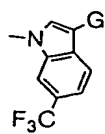
J16



J17



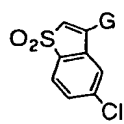
J18



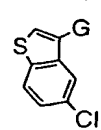
J19



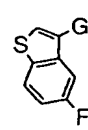
J20



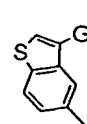
J21



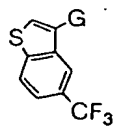
J22



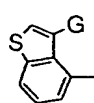
J23



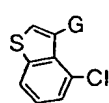
J24



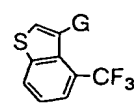
J25



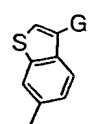
J26



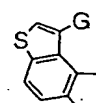
J27



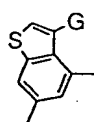
J28



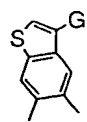
J29



J30



J31



J32

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1	H	H	A1	J1	S
2	H	H	A1	J2	S
3	H	H	A1	J3	S
4	H	H	A1	J4	S
5	H	H	A1	J5	S
6	H	H	A1	J6	S
7	H	H	A1	J7	S
8	H	H	A1	J8	S
9	H	H	A1	J9	S
10	H	H	A1	J10	S
11	H	H	A1	J11	S
12	H	H	A1	J12	S
13	H	H	A1	J13	S
14	H	H	A1	J14	S
15	H	H	A1	J15	S
16	H	H	A1	J16	S
17	H	H	A1	J17	S
18	H	H	A1	J18	S
19	H	H	A1	J19	S
20	H	H	A1	J20	S
21	H	H	A1	J21	S
22	H	H	A1	J22	S
23	H	H	A1	J23	S
24	H	H	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
25	H	H	A1	J25	S
26	H	H	A1	J26	S
27	H	H	A1	J27	S
28	H	H	A1	J28	S
29	H	H	A1	J29	S
30	H	H	A1	J30	S
31	H	H	A1	J31	S
32	H	H	A1	J32	S
33	H	H	A2	J1	S
34	H	H	A2	J2	S
35	H	H	A2	J3	S
36	H	H	A2	J4	S
37	H	H	A2	J5	S
38	H	H	A2	J6	S
39	H	H	A2	J7	S
40	H	H	A2	J8	S
41	H	H	A2	J9	S
42	H	H	A2	J10	S
43	H	H	A2	J11	S
44	H	H	A2	J12	S
45	H	H	A2	J13	S
46	H	H	A2	J14	S
47	H	H	A2	J15	S
48	H	H	A2	J16	S



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
49	H	H	A2	J17	S
50	H	H	A2	J18	S
51	H	H	A2	J19	S
52	H	H	A2	J20	S
53	H	H	A2	J21	S
54	H	H	A2	J22	S
55	H	H	A2	J23	S
56	H	H	A2	J24	S
57	H	H	A2	J25	S
58	H	H	A2	J26	S
59	H	H	A2	J27	S
60	H	H	A2	J28	S
61	H	H	A2	J29	S
62	H	H	A2	J30	S
63	H	H	A2	J31	S
64	H	H	A2	J32	S
65	H	H	A3	J1	S
66	H	H	A3	J2	S
67	H	H	A3	J3	S
68	H	H	A3	J4	S
69	H	H	A3	J5	S
70	H	H	A3	J6	S
71	H	H	A3	J7	S
72	H	H	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
73	H	H	A3	J9	S
74	H	H	A3	J10	S
75	H	H	A3	J11	S
76	H	H	A3	J12	S
77	H	H	A3	J13	S
78	H	H	A3	J14	S
79	H	H	A3	J15	S
80	H	H	A3	J16	S
81	H	H	A3	J17	S
82	H	H	A3	J18	S
83	H	H	A3	J19	S
84	H	H	A3	J20	S
85	H	H	A3	J21	S
86	H	H	A3	J22	S
87	H	H	A3	J23	S
88	H	H	A3	J24	S
89	H	H	A3	J25	S
90	H	H	A3	J26	S
91	H	H	A3	J27	S
92	H	H	A3	J28	S
93	H	H	A3	J29	S
94	H	H	A3	J30	S
95	H	H	A3	J31	S
96	H	H	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
97	MeO	H	A1	J1	S
98	MeO	H	A1	J2	S
99	MeO	H	A1	J3	S
100	MeO	H	A1	J4	S
101	MeO	H	A1	J5	S
102	MeO	H	A1	J6	S
103	MeO	H	A1	J7	S
104	MeO	H	A1	J8	S
105	MeO	H	A1	J9	S
106	MeO	H	A1	J10	S
107	MeO	H	A1	J11	S
108	MeO	H	A1	J12	S
109	MeO	H	A1	J13	S
110	MeO	H	A1	J14	S
111	MeO	H	A1	J15	S
112	MeO	H	A1	J16	S
113	MeO	H	A1	J17	S
114	MeO	H	A1	J18	S
115	MeO	H	A1	J19	S
116	MeO	H	A1	J20	S
117	MeO	H	A1	J21	S
118	MeO	H	A1	J22	S
119	MeO	H	A1	J23	S
120	MeO	H	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
121	MeO	H	A1	J25	S
122	MeO	H	A1	J26	S
123	MeO	H	A1	J27	S
124	MeO	H	A1	J28	S
125	MeO	H	A1	J29	S
126	MeO	H	A1	J30	S
127	MeO	H	A1	J31	S
128	MeO	H	A1	J32	S
129	MeO	H	A2	J1	S
130	MeO	H	A2	J2	S
131	MeO	H	A2	J3	S
132	MeO	H	A2	J4	S
133	MeO	H	A2	J5	S
134	MeO	H	A2	J6	S
135	MeO	H	A2	J7	S
136	MeO	H	A2	J8	S
137	MeO	H	A2	J9	S
138	MeO	H	A2	J10	S
139	MeO	H	A2	J11	S
140	MeO	H	A2	J12	S
141	MeO	H	A2	J13	S
142	MeO	H	A2	J14	S
143	MeO	H	A2	J15	S
144	MeO	H	A2	J16	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
145	MeO	H	A2	J17	S
146	MeO	H	A2	J18	S
147	MeO	H	A2	J19	S
148	MeO	H	A2	J20	S
149	MeO	H	A2	J21	S
150	MeO	H	A2	J22	S
151	MeO	H	A2	J23	S
152	MeO	H	A2	J24	S
153	MeO	H	A2	J25	S
154	MeO	H	A2	J26	S
155	MeO	H	A2	J27	S
156	MeO	H	A2	J28	S
157	MeO	H	A2	J29	S
158	MeO	H	A2	J30	S
159	MeO	H	A2	J31	S
160	MeO	H	A2	J32	S
161	MeO	H	A3	J1	S
162	MeO	H	A3	J2	S
163	MeO	H	A3	J3	S
164	MeO	H	A3	J4	S
165	MeO	H	A3	J5	S
166	MeO	H	A3	J6	S
167	MeO	H	A3	J7	S
168	MeO	H	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
169	MeO	H	A3	J9	S
170	MeO	H	A3	J10	S
171	MeO	H	A3	J11	S
172	MeO	H	A3	J12	S
173	MeO	H	A3	J13	S
174	MeO	H	A3	J14	S
175	MeO	H	A3	J15	S
176	MeO	H	A3	J16	S
177	MeO	H	A3	J17	S
178	MeO	H	A3	J18	S
179	MeO	H	A3	J19	S
180	MeO	H	A3	J20	S
181	MeO	H	A3	J21	S
182	MeO	H	A3	J22	S
183	MeO	H	A3	J23	S
184	MeO	H	A3	J24	S
185	MeO	H	A3	J25	S
186	MeO	H	A3	J26	S
187	MeO	H	A3	J27	S
188	MeO	H	A3	J28	S
189	MeO	H	A3	J29	S
190	MeO	H	A3	J30	S
191	MeO	H	A3	J31	S
192	MeO	H	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
193	CN	H	A1	J1	S
194	CN	H	A1	J2	S
195	CN	H	A1	J3	S
196	CN	H	A1	J4	S
197	CN	H	A1	J5	S
198	CN	H	A1	J6	S
199	CN	H	A1	J7	S
200	CN	H	A1	J8	S
201	CN	H	A1	J9	S
202	CN	H	A1	J10	S
203	CN	H	A1	J11	S
204	CN	H	A1	J12	S
205	CN	H	A1	J13	S
206	CN	H	A1	J14	S
207	CN	H	A1	J15	S
208	CN	H	A1	J16	S
209	CN	H	A1	J17	S
210	CN	H	A1	J18	S
211	CN	H	A1	J19	S
212	CN	H	A1	J20	S
213	CN	H	A1	J21	S
214	CN	H	A1	J22	S
215	CN	H	A1	J23	S
216	CN	H	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
217	CN	H	A1	J25	S
218	CN	H	A1	J26	S
219	CN	H	A1	J27	S
220	CN	H	A1	J28	S
221	CN	H	A1	J29	S
222	CN	H	A1	J30	S
223	CN	H	A1	J31	S
224	CN	H	A1	J32	S
225	CN	H	A2	J1	S
226	CN	H	A2	J2	S
227	CN	H	A2	J3	S
228	CN	H	A2	J4	S
229	CN	H	A2	J5	S
230	CN	H	A2	J6	S
231	CN	H	A2	J7	S
232	CN	H	A2	J8	S
233	CN	H	A2	J9	S
234	CN	H	A2	J10	S
235	CN	H	A2	J11	S
236	CN	H	A2	J12	S
237	CN	H	A2	J13	S
238	CN	H	A2	J14	S
239	CN	H	A2	J15	S
240	CN	H	A2	J16	S



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
241	CN	H	A2	J17	S
242	CN	H	A2	J18	S
243	CN	H	A2	J19	S
244	CN	H	A2	J20	S
245	CN	H	A2	J21	S
246	CN	H	A2	J22	S
247	CN	H	A2	J23	S
248	CN	H	A2	J24	S
249	CN	H	A2	J25	S
250	CN	H	A2	J26	S
251	CN	H	A2	J27	S
252	CN	H	A2	J28	S
253	CN	H	A2	J29	S
254	CN	H	A2	J30	S
255	CN	H	A2	J31	S
256	CN	H	A2	J32	S
257	CN	H	A3	J1	S
258	CN	H	A3	J2	S
259	CN	H	A3	J3	S
260	CN	H	A3	J4	S
261	CN	H	A3	J5	S
262	CN	H	A3	J6	S
263	CN	H	A3	J7	S
264	CN	H	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
265	CN	H	A3	J9	S
266	CN	H	A3	J10	S
267	CN	H	A3	J11	S
268	CN	H	A3	J12	S
269	CN	H	A3	J13	S
270	CN	H	A3	J14	S
271	CN	H	A3	J15	S
272	CN	H	A3	J16	S
273	CN	H	A3	J17	S
274	CN	H	A3	J18	S
275	CN	H	A3	J19	S
276	CN	H	A3	J20	S
277	CN	H	A3	J21	S
278	CN	H	A3	J22	S
279	CN	H	A3	J23	S
280	CN	H	A3	J24	S
281	CN	H	A3	J25	S
282	CN	H	A3	J26	S
283	CN	H	A3	J27	S
284	CN	H	A3	J28	S
285	CN	H	A3	J29	S
286	CN	H	A3	J30	S
287	CN	H	A3	J31	S
288	CN	H	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
289	Me	H	A1	J1	S
290	Me	H	A1	J2	S
291	Me	H	A1	J3	S
292	Me	H	A1	J4	S
293	Me	H	A1	J5	S
294	Me	H	A1	J6	S
295	Me	H	A1	J7	S
296	Me	H	A1	J8	S
297	Me	H	A1	J9	S
298	Me	H	A1	J10	S
299	Me	H	A1	J11	S
300	Me	H	A1	J12	S
301	Me	H	A1	J13	S
302	Me	H	A1	J14	S
303	Me	H	A1	J15	S
304	Me	H	A1	J16	S
305	Me	H	A1	J17	S
306	Me	H	A1	J18	S
307	Me	H	A1	J19	S
308	Me	H	A1	J20	S
309	Me	H	A1	J21	S
310	Me	H	A1	J22	S
311	Me	H	A1	J23	S
312	Me	H	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
313	Me	H	A1	J25	S
314	Me	H	A1	J26	S
315	Me	H	A1	J27	S
316	Me	H	A1	J28	S
317	Me	H	A1	J29	S
318	Me	H	A1	J30	S
319	Me	H	A1	J31	S
320	Me	H	A1	J32	S
321	Me	H	A2	J1	S
322	Me	H	A2	J2	S
323	Me	H	A2	J3	S
324	Me	H	A2	J4	S
325	Me	H	A2	J5	S
326	Me	H	A2	J6	S
327	Me	H	A2	J7	S
328	Me	H	A2	J8	S
329	Me	H	A2	J9	S
330	Me	H	A2	J10	S
331	Me	H	A2	J11	S
332	Me	H	A2	J12	S
333	Me	H	A2	J13	S
334	Me	H	A2	J14	S
335	Me	H	A2	J15	S
336	Me	H	A2	J16	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
337	Me	H	A2	J17	S
338	Me	H	A2	J18	S
339	Me	H	A2	J19	S
340	Me	H	A2	J20	S
341	Me	H	A2	J21	S
342	Me	H	A2	J22	S
343	Me	H	A2	J23	S
344	Me	H	A2	J24	S
345	Me	H	A2	J25	S
346	Me	H	A2	J26	S
347	Me	H	A2	J27	S
348	Me	H	A2	J28	S
349	Me	H	A2	J29	S
350	Me	H	A2	J30	S
351	Me	H	A2	J31	S
352	Me	H	A2	J32	S
353	Me	H	A3	J1	S
354	Me	H	A3	J2	S
355	Me	H	A3	J3	S
356	Me	H	A3	J4	S
357	Me	H	A3	J5	S
358	Me	H	A3	J6	S
359	Me	H	A3	J7	S
360	Me	H	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
361	Me	H	A3	J9	S
362	Me	H	A3	J10	S
363	Me	H	A3	J11	S
364	Me	H	A3	J12	S
365	Me	H	A3	J13	S
366	Me	H	A3	J14	S
367	Me	H	A3	J15	S
368	Me	H	A3	J16	S
369	Me	H	A3	J17	S
370	Me	H	A3	J18	S
371	Me	H	A3	J19	S
372	Me	H	A3	J20	S
373	Me	H	A3	J21	S
374	Me	H	A3	J22	S
375	Me	H	A3	J23	S
376	Me	H	A3	J24	S
377	Me	H	A3	J25	S
378	Me	H	A3	J26	S
379	Me	H	A3	J27	S
380	Me	H	A3	J28	S
381	Me	H	A3	J29	S
382	Me	H	A3	J30	S
383	Me	H	A3	J31	S
384	Me	H	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
385	H	Me	A1	J1	S
386	H	Me	A1	J2	S
387	H	Me	A1	J3	S
388	H	Me	A1	J4	S
389	H	Me	A1	J5	S
390	H	Me	A1	J6	S
391	H	Me	A1	J7	S
392	H	Me	A1	J8	S
393	H	Me	A1	J9	S
394	H	Me	A1	J10	S
395	H	Me	A1	J11	S
396	H	Me	A1	J12	S
397	H	Me	A1	J13	S
398	H	Me	A1	J14	S
399	H	Me	A1	J15	S
400	H	Me	A1	J16	S
401	H	Me	A1	J17	S
402	H	Me	A1	J18	S
403	H	Me	A1	J19	S
404	H	Me	A1	J20	S
405	H	Me	A1	J21	S
406	H	Me	A1	J22	S
407	H	Me	A1	J23	S
408	H	Me	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
409	H	Me	A1	J25	S
410	H	Me	A1	J26	S
411	H	Me	A1	J27	S
412	H	Me	A1	J28	S
413	H	Me	A1	J29	S
414	H	Me	A1	J30	S
415	H	Me	A1	J31	S
416	H	Me	A1	J32	S
417	H	Me	A2	J1	S
418	H	Me	A2	J2	S
419	H	Me	A2	J3	S
420	H	Me	A2	J4	S
421	H	Me	A2	J5	S
422	H	Me	A2	J6	S
423	H	Me	A2	J7	S
424	H	Me	A2	J8	S
425	H	Me	A2	J9	S
426	H	Me	A2	J10	S
427	H	Me	A2	J11	S
428	H	Me	A2	J12	S
429	H	Me	A2	J13	S
430	H	Me	A2	J14	S
431	H	Me	A2	J15	S
432	H	Me	A2	J16	S



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
433	H	Me	A2	J17	S
434	H	Me	A2	J18	S
435	H	Me	A2	J19	S
436	H	Me	A2	J20	S
437	H	Me	A2	J21	S
438	H	Me	A2	J22	S
439	H	Me	A2	J23	S
440	H	Me	A2	J24	S
441	H	Me	A3	J25	S
442	H	Me	A2	J26	S
443	H	Me	A2	J27	S
444	H	Me	A2	J28	S
445	H	Me	A2	J29	S
446	H	Me	A2	J30	S
447	H	Me	A2	J31	S
448	H	Me	A2	J32	S
449	H	Me	A3	J1	S
450	H	Me	A3	J2	S
451	H	Me	A3	J3	S
452	H	Me	A3	J4	S
453	H	Me	A3	J5	S
454	H	Me	A3	J6	S
455	H	Me	A3	J7	S
456	H	Me	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
457	H	Me	A3	J9	S
458	H	Me	A3	J10	S
459	H	Me	A3	J11	S
460	H	Me	A3	J12	S
461	H	Me	A3	J13	S
462	H	Me	A3	J14	S
463	H	Me	A3	J15	S
464	H	Me	A3	J16	S
465	H	Me	A3	J17	S
466	H	Me	A3	J18	S
467	H	Me	A3	J19	S
468	H	Me	A3	J20	S
469	H	Me	A3	J21	S
470	H	Me	A3	J22	S
471	H	Me	A3	J23	S
472	H	Me	A3	J24	S
473	H	Me	A3	J25	S
474	H	Me	A3	J26	S
475	H	Me	A3	J27	S
476	H	Me	A3	J28	S
477	H	Me	A3	J29	S
478	H	Me	A3	J30	S
479	H	Me	A3	J31	S
480	H	Me	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
481	Me	Me	A1	J1	S
482	Me	Me	A1	J2	S
483	Me	Me	A1	J3	S
484	Me	Me	A1	J4	S
485	Me	Me	A1	J5	S
486	Me	Me	A1	J6	S
487	Me	Me	A1	J7	S
488	Me	Me	A1	J8	S
489	Me	Me	A1	J9	S
490	Me	Me	A1	J10	S
491	Me	Me	A1	J11	S
492	Me	Me	A1	J12	S
493	Me	Me	A1	J13	S
494	Me	Me	A1	J14	S
495	Me	Me	A1	J15	S
496	Me	Me	A1	J16	S
497	Me	Me	A1	J17	S
498	Me	Me	A1	J18	S
499	Me	Me	A1	J19	S
500	Me	Me	A1	J20	S
501	Me	Me	A1	J21	S
502	Me	Me	A1	J22	S
503	Me	Me	A1	J23	S
504	Me	Me	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
505	Me	Me	A1	J25	S
506	Me	Me	A1	J26	S
507	Me	Me	A1	J27	S
508	Me	Me	A1	J28	S
509	Me	Me	A1	J29	S
510	Me	Me	A1	J30	S
511	Me	Me	A1	J31	S
512	Me	Me	A1	J32	S
513	Me	Me	A2	J1	S
514	Me	Me	A2	J2	S
515	Me	Me	A2	J3	S
516	Me	Me	A2	J4	S
517	Me	Me	A2	J5	S
518	Me	Me	A2	J6	S
519	Me	Me	A2	J7	S
520	Me	Me	A2	J8	S
521	Me	Me	A2	J9	S
522	Me	Me	A2	J10	S
523	Me	Me	A2	J11	S
524	Me	Me	A2	J12	S
525	Me	Me	A2	J13	S
526	Me	Me	A2	J14	S
257	Me	Me	A2	J15	S
528	Me	Me	A2	J16	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
529	Me	Me	A2	J17	S
530	Me	Me	A2	J18	S
531	Me	Me	A2	J19	S
532	Me	Me	A2	J20	S
533	Me	Me	A2	J21	S
534	Me	Me	A2	J22	S
535	Me	Me	A2	J23	S
536	Me	Me	A2	J24	S
537	Me	Me	A2	J25	S
538	Me	Me	A2	J26	S
539	Me	Me	A2	J27	S
540	Me	Me	A2	J28	S
541	Me	Me	A2	J29	S
542	Me	Me	A2	J30	S
543	Me	Me	A2	J31	S
544	Me	Me	A2	J32	S
545	Me	Me	A3	J1	S
546	Me	Me	A3	J2	S
547	Me	Me	A3	J3	S
548	Me	Me	A3	J4	S
549	Me	Me	A3	J5	S
550	Me	Me	A3	J6	S
551	Me	Me	A3	J7	S
552	Me	Me	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
553	Me	Me	A3	J9	S
554	Me	Me	A3	J10	S
555	Me	Me	A3	J11	S
556	Me	Me	A3	J12	S
557	Me	Me	A3	J13	S
558	Me	Me	A3	J14	S
559	Me	Me	A3	J15	S
560	Me	Me	A3	J16	S
561	Me	Me	A3	J17	S
562	Me	Me	A3	J18	S
563	Me	Me	A3	J19	S
564	Me	Me	A3	J20	S
565	Me	Me	A3	J21	S
566	Me	Me	A3	J22	S
567	Me	Me	A3	J23	S
568	Me	Me	A3	J24	S
569	Me	Me	A3	J25	S
570	Me	Me	A3	J26	S
571	Me	Me	A3	J27	S
572	Me	Me	A3	J28	S
573	Me	Me	A3	J29	S
574	Me	Me	A3	J30	S
575	Me	Me	A3	J31	S
576	Me	Me	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
577	C1	C1	A1	J1	S
578	C1	C1	A1	J2	S
579	C1	C1	A1	J3	S
580	C1	C1	A1	J4	S
581	C1	C1	A1	J5	S
582	C1	C1	A1	J6	S
583	C1	C1	A1	J7	S
584	C1	C1	A1	J8	S
585	C1	C1	A1	J9	S
586	C1	C1	A1	J10	S
587	C1	C1	A1	J11	S
588	C1	C1	A1	J12	S
589	C1	C1	A1	J13	S
590	C1	C1	A1	J14	S
591	C1	C1	A1	J15	S
592	C1	C1	A1	J16	S
593	C1	C1	A1	J17	S
594	C1	C1	A1	J18	S
595	C1	C1	A1	J19	S
596	C1	C1	A1	J20	S
597	C1	C1	A1	J21	S
598	C1	C1	A1	J22	S
599	C1	C1	A1	J23	S
600	C1	C1	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
601	C1	C1	A1	J25	S
602	C1	C1	A1	J26	S
603	C1	C1	A1	J27	S
604	C1	C1	A1	J28	S
605	C1	C1	A1	J29	S
606	C1	C1	A1	J30	S
607	C1	C1	A1	J31	S
608	C1	C1	A1	J32	S
609	C1	C1	A2	J1	S
610	C1	C1	A2	J2	S
611	C1	C1	A2	J3	S
612	C1	C1	A2	J4	S
613	C1	C1	A2	J5	S
614	C1	C1	A2	J6	S
615	C1	C1	A2	J7	S
616	C1	C1	A2	J8	S
617	C1	C1	A2	J9	S
618	C1	C1	A2	J10	S
619	C1	C1	A2	J11	S
620	C1	C1	A2	J12	S
621	C1	C1	A2	J13	S
622	C1	C1	A2	J14	S
623	C1	C1	A2	J15	S
624	C1	C1	A2	J16	S



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
625	C1	C1	A2	J17	S
626	C1	C1	A2	J18	S
627	C1	C1	A2	J19	S
628	C1	C1	A2	J20	S
629	C1	C1	A2	J21	S
630	C1	C1	A2	J22	S
631	C1	C1	A2	J23	S
632	C1	C1	A2	J24	S
633	C1	C1	A2	J25	S
634	C1	C1	A2	J26	S
635	C1	C1	A2	J27	S
636	C1	C1	A2	J28	S
637	C1	C1	A2	J29	S
638	C1	C1	A2	J30	S
639	C1	C1	A2	J31	S
640	C1	C1	A2	J32	S
641	C1	C1	A3	J1	S
642	C1	C1	A3	J2	S
643	C1	C1	A3	J3	S
644	C1	C1	A3	J4	S
645	C1	C1	A3	J5	S
646	C1	C1	A3	J6	S
647	C1	C1	A3	J7	S
648	C1	C1	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
649	C1	C1	A3	J9	S
650	C1	C1	A3	J10	S
651	C1	C1	A3	J11	S
652	C1	C1	A3	J12	S
653	C1	C1	A3	J13	S
654	C1	C1	A3	J14	S
655	C1	C1	A3	J15	S
656	C1	C1	A3	J16	S
657	C1	C1	A3	J17	S
658	C1	C1	A3	J18	S
659	C1	C1	A3	J19	S
660	C1	C1	A3	J20	S
661	C1	C1	A3	J21	S
662	C1	C1	A3	J22	S
663	C1	C1	A3	J23	S
664	C1	C1	A3	J24	S
665	C1	C1	A3	J25	S
666	C1	C1	A3	J26	S
667	C1	C1	A3	J27	S
668	C1	C1	A3	J28	S
669	C1	C1	A3	J29	S
670	C1	C1	A3	J30	S
671	C1	C1	A3	J31	S
672	C1	C1	A3	J32	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
673	H	H	A1	J1	-
674	H	H	A1	J2	-
675	H	H	A1	J3	-
676	H	H	A1	J4	-
677	H	H	A1	J5	-
678	H	H	A1	J6	-
679	H	H	A1	J7	-
680	H	H	A1	J8	-
681	H	H	A1	J9	-
682	H	H	A1	J10	-
683	H	H	A1	J11	-
684	H	H	A1	J12	-
685	H	H	A1	J13	-
686	H	H	A1	J14	-
687	H	H	A1	J15	-
688	H	H	A1	J16	-
689	H	H	A1	J17	-
690	H	H	A1	J18	-
691	H	H	A1	J19	-
692	H	H	A1	J20	-
693	H	H	A1	J21	-
694	H	H	A1	J22	-
695	H	H	A1	J23	-
696	H	H	A1	J24	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
697	H	H	A1	J25	-
698	H	H	A1	J26	-
699	H	H	A1	J27	-
700	H	H	A1	J28	-
701	H	H	A1	J29	-
702	H	H	A1	J30	-
703	H	H	A1	J31	-
704	H	H	A1	J32	-
705	H	H	A2	J1	-
706	H	H	A2	J2	-
707	H	H	A2	J3	-
708	H	H	A2	J4	-
709	H	H	A2	J5	-
710	H	H	A2	J6	-
711	H	H	A2	J7	-
712	H	H	A2	J8	-
713	H	H	A2	J9	-
714	H	H	A2	J10	-
715	H	H	A2	J11	-
716	H	H	A2	J12	-
717	H	H	A2	J13	-
718	H	H	A2	J14	-
719	H	H	A2	J15	-
720	H	H	A2	J16	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
721	H	H	A2	J17	-
722	H	H	A2	J18	-
723	H	H	A2	J19	-
724	H	H	A2	J20	-
725	H	H	A2	J21	-
726	H	H	A2	J22	-
727	H	H	A2	J23	-
728	H	H	A2	J24	-
729	H	H	A2	J25	-
730	H	H	A2	J26	-
731	H	H	A2	J27	-
732	H	H	A2	J28	-
733	H	H	A2	J29	-
734	H	H	A2	J30	-
735	H	H	A2	J31	-
736	H	H	A2	J32	-
737	H	H	A3	J1	-
738	H	H	A3	J2	-
739	H	H	A3	J3	-
740	H	H	A3	J4	-
741	H	H	A3	J5	-
742	H	H	A3	J6	-
743	H	H	A3	J7	-
744	H	H	A3	J8	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
745	H	H	A3	J9	-
746	H	H	A3	J10	-
747	H	H	A3	J11	-
748	H	H	A3	J12	-
749	H	H	A3	J13	-
750	H	H	A3	J14	-
751	H	H	A3	J15	-
752	H	H	A3	J16	-
753	H	H	A3	J17	-
754	H	H	A3	J18	-
755	H	H	A3	J19	-
756	H	H	A3	J20	-
757	H	H	A3	J21	-
758	H	H	A3	J22	-
759	H	H	A3	J23	-
760	H	H	A3	J24	-
761	H	H	A3	J25	-
762	H	H	A3	J26	-
763	H	H	A3	J27	-
764	H	H	A3	J28	-
765	H	H	A3	J29	-
766	H	H	A3	J30	-
767	H	H	A3	J31	-
768	H	H	A3	J32	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
769	MeO	H	A1	J1	-
770	MeO	H	A1	J2	-
771	MeO	H	A1	J3	-
772	MeO	H	A1	J4	-
773	MeO	H	A1	J5	-
774	MeO	H	A1	J6	-
775	MeO	H	A1	J7	-
776	MeO	H	A1	J8	-
777	MeO	H	A1	J9	-
778	MeO	H	A1	J10	-
779	MeO	H	A1	J11	-
780	MeO	H	A1	J12	-
781	MeO	H	A1	J13	-
782	MeO	H	A1	J14	-
783	MeO	H	A1	J15	-
784	MeO	H	A1	J16	-
785	MeO	H	A1	J17	-
786	MeO	H	A1	J18	-
787	MeO	H	A1	J19	-
788	MeO	H	A1	J20	-
789	MeO	H	A1	J21	-
790	MeO	H	A1	J22	-
791	MeO	H	A1	J23	-
792	MeO	H	A1	J24	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
793	MeO	H	A1	J25	-
794	MeO	H	A1	J26	-
795	MeO	H	A1	J27	-
796	MeO	H	A1	J28	-
797	MeO	H	A1	J29	-
798	MeO	H	A1	J30	-
799	MeO	H	A1	J31	-
800	MeO	H	A1	J32	-
801	MeO	H	A2	J1	-
802	MeO	H	A2	J2	-
803	MeO	H	A2	J3	-
804	MeO	H	A2	J4	-
805	MeO	H	A2	J5	-
806	MeO	H	A2	J6	-
807	MeO	H	A2	J7	-
808	MeO	H	A2	J8	-
809	MeO	H	A2	J9	-
810	MeO	H	A2	J10	-
811	MeO	H	A2	J11	-
812	MeO	H	A2	J12	-
813	MeO	H	A2	J13	-
814	MeO	H	A2	J14	-
815	MeO	H	A2	J15	-
816	MeO	H	A2	J16	-



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
817	MeO	H	A2	J17	-
818	MeO	H	A2	J18	-
819	MeO	H	A2	J19	-
820	MeO	H	A2	J20	-
821	MeO	H	A2	J21	-
822	MeO	H	A2	J22	-
823	MeO	H	A2	J23	-
824	MeO	H	A2	J24	-
825	MeO	H	A2	J25	-
826	MeO	H	A2	J26	-
827	MeO	H	A2	J27	-
828	MeO	H	A2	J28	-
829	MeO	H	A2	J29	-
830	MeO	H	A2	J30	-
931	MeO	H	A2	J31	-
832	MeO	H	A2	J32	-
833	MeO	H	A3	J1	-
834	MeO	H	A3	J2	-
835	MeO	H	A3	J3	-
836	MeO	H	A3	J4	-
837	MeO	H	A3	J5	-
838	MeO	H	A3	J6	-
839	MeO	H	A3	J7	-
840	MeO	H	A3	J8	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
841	MeO	H	A3	J9	-
842	MeO	H	A3	J10	-
843	MeO	H	A3	J11	-
844	MeO	H	A3	J12	-
845	MeO	H	A3	J13	-
846	MeO	H	A3	J14	-
847	MeO	H	A3	J15	-
848	MeO	H	A3	J16	-
849	MeO	H	A3	J17	-
850	MeO	H	A3	J18	-
851	MeO	H	A3	J19	-
852	MeO	H	A3	J20	-
853	MeO	H	A3	J21	-
854	MeO	H	A3	J22	-
855	MeO	H	A3	J23	-
856	MeO	H	A3	J24	-
857	MeO	H	A3	J25	-
858	MeO	H	A3	J26	-
859	MeO	H	A3	J27	-
860	MeO	H	A3	J28	-
861	MeO	H	A3	J29	-
862	MeO	H	A3	J30	-
863	MeO	H	A3	J31	-
864	MeO	H	A3	J32	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
865	CN	H	A1	J1	-
866	CN	H	A1	J2	-
867	CN	H	A1	J3	-
868	CN	H	A1	J4	-
869	CN	H	A1	J5	-
870	CN	H	A1	J6	-
871	CN	H	A1	J7	-
872	CN	H	A1	J8	-
873	CN	H	A1	J9	-
874	CN	H	A1	J10	-
875	CN	H	A1	J11	-
876	CN	H	A1	J12	-
877	CN	H	A1	J13	-
878	CN	H	A1	J14	-
879	CN	H	A1	J15	-
880	CN	H	A1	J16	-
881	CN	H	A1	J17	-
882	CN	H	A1	J18	-
883	CN	H	A1	J19	-
884	CN	H	A1	J20	-
885	CN	H	A1	J21	-
886	CN	H	A1	J22	-
887	CN	H	A1	J23	-
888	CN	H	A1	J24	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
889	CN	H	A1	J25	-
890	CN	H	A1	J26	-
891	CN	H	A1	J27	-
892	CN	H	A1	J28	-
893	CN	H	A1	J29	-
894	CN	H	A1	J30	-
895	CN	H	A1	J31	-
896	CN	H	A1	J32	-
897	CN	H	A2	J1	-
898	CN	H	A2	J2	-
899	CN	H	A2	J3	-
900	CN	H	A2	J4	-
901	CN	H	A2	J5	-
902	CN	H	A2	J6	-
903	CN	H	A2	J7	-
904	CN	H	A2	J8	-
905	CN	H	A2	J9	-
906	CN	H	A2	J10	-
907	CN	H	A2	J11	-
908	CN	H	A2	J12	-
909	CN	H	A2	J13	-
910	CN	H	A2	J14	-
911	CN	H	A2	J15	-
912	CN	H	A2	J16	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
913	CN	H	A2	J17	-
914	CN	H	A2	J18	-
915	CN	H	A2	J19	-
916	CN	H	A2	J20	-
917	CN	H	A2	J21	-
918	CN	H	A2	J22	-
919	CN	H	A2	J23	-
920	CN	H	A2	J24	-
921	CN	H	A3	J25	-
922	CN	H	A2	J26	-
923	CN	H	A2	J27	-
924	CN	H	A2	J28	-
925	CN	H	A2	J29	-
926	CN	H	A2	J30	-
927	CN	H	A2	J31	-
928	CN	H	A2	J32	-
929	CN	H	A3	J1	-
930	CN	H	A3	J2	-
931	CN	H	A3	J3	-
932	CN	H	A3	J4	-
933	CN	H	A3	J5	-
934	CN	H	A3	J6	-
935	CN	H	A3	J7	-
936	CN	H	A3	J8	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
937	CN	H	A3	J9	-
938	CN	H	A3	J10	-
939	CN	H	A3	J11	-
940	CN	H	A3	J12	-
941	CN	H	A3	J13	-
942	CN	H	A3	J14	-
943	CN	H	A3	J15	-
944	CN	H	A3	J16	-
945	CN	H	A3	J17	-
946	CN	H	A3	J18	-
947	CN	H	A3	J19	-
948	CN	H	A3	J20	-
949	CN	H	A3	J21	-
950	CN	H	A3	J22	-
951	CN	H	A3	J23	-
952	CN	H	A3	J24	-
953	CN	H	A3	J25	-
954	CN	H	A3	J26	-
955	CN	H	A3	J27	-
956	CN	H	A3	J28	-
957	CN	H	A3	J29	-
958	CN	H	A3	J30	-
959	CN	H	A3	J31	-
960	CN	H	A3	J32	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
961	Me	Me	A1	J1	-
962	Me	Me	A1	J2	-
963	Me	Me	A1	J3	-
964	Me	Me	A1	J4	-
965	Me	Me	A1	J5	-
966	Me	Me	A1	J6	-
967	Me	Me	A1	J7	-
968	Me	Me	A1	J8	-
969	Me	Me	A1	J9	-
970	Me	Me	A1	J10	-
971	Me	Me	A1	J11	-
972	Me	Me	A1	J12	-
973	Me	Me	A1	J13	-
974	Me	Me	A1	J14	-
975	Me	Me	A1	J15	-
976	Me	Me	A1	J16	-
977	Me	Me	A1	J17	-
978	Me	Me	A1	J18	-
979	Me	Me	A1	J19	-
980	Me	Me	A1	J20	-
981	Me	Me	A1	J21	-
982	Me	Me	A1	J22	-
983	Me	Me	A1	J23	-
984	Me	Me	A1	J24	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
985	Me	Me	A1	J25	-
986	Me	Me	A1	J26	-
987	Me	Me	A1	J27	-
988	Me	Me	A1	J28	-
989	Me	Me	A1	J29	-
990	Me	Me	A1	J30	-
991	Me	Me	A1	J31	-
992	Me	Me	A1	J32	-
993	Me	Me	A2	J1	-
994	Me	Me	A2	J2	-
995	Me	Me	A2	J3	-
996	Me	Me	A2	J4	-
997	Me	Me	A2	J5	-
998	Me	Me	A2	J6	-
999	Me	Me	A2	J7	-
1000	Me	Me	A2	J8	-
1001	Me	Me	A2	J9	-
1002	Me	Me	A2	J10	-
1003	Me	Me	A2	J11	-
1004	Me	Me	A2	J12	-
1005	Me	Me	A2	J13	-
1006	Me	Me	A2	J14	-
1007	Me	Me	A2	J15	-
1008	Me	Me	A2	J16	-



Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1009	Me	Me	A2	J17	-
1010	Me	Me	A2	J18	-
1011	Me	Me	A2	J19	-
1012	Me	Me	A2	J20	-
1013	Me	Me	A2	J21	-
1014	Me	Me	A2	J22	-
1015	Me	Me	A2	J23	-
1016	Me	Me	A2	J24	-
1017	Me	Me	A3	J25	-
1018	Me	Me	A2	J26	-
1019	Me	Me	A2	J27	-
1020	Me	Me	A2	J28	-
1021	Me	Me	A2	J29	-
1022	Me	Me	A2	J30	-
1023	Me	Me	A2	J31	-
1024	Me	Me	A2	J32	-
1025	Me	Me	A3	J1	-
1026	Me	Me	A3	J2	-
1027	Me	Me	A3	J3	-
1028	Me	Me	A3	J4	-
1029	Me	Me	A3	J5	-
1030	Me	Me	A3	J6	-
1031	Me	Me	A3	J7	-
1032	Me	Me	A3	J8	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1033	Me	Me	A3	J9	-
1034	Me	Me	A3	J10	-
1035	Me	Me	A3	J11	-
1036	Me	Me	A3	J12	-
1037	Me	Me	A3	J13	-
1038	Me	Me	A3	J14	-
1039	Me	Me	A3	J15	-
1040	Me	Me	A3	J16	-
1041	Me	Me	A3	J17	-
1042	Me	Me	A3	J18	-
1043	Me	Me	A3	J19	-
1044	Me	Me	A3	J20	-
1045	Me	Me	A3	J21	-
1046	Me	Me	A3	J22	-
1047	Me	Me	A3	J23	-
1048	Me	Me	A3	J24	-
1049	Me	Me	A3	J25	-
1050	Me	Me	A3	J26	-
1051	Me	Me	A3	J27	-
1052	Me	Me	A3	J28	-
1053	Me	Me	A3	J29	-
1054	Me	Me	A3	J30	-
1055	Me	Me	A3	J31	-
1056	Me	Me	A3	J32	-

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1057	H	MeO	A1	J1	S
1058	H	MeO	A1	J2	S
1059	H	MeO	A1	J3	S
1060	H	MeO	A1	J4	S
1061	H	MeO	A1	J5	S
1062	H	MeO	A1	J6	S
1063	H	MeO	A1	J7	S
1064	H	MeO	A1	J8	S
1065	H	MeO	A1	J9	S
1066	H	MeO	A1	J10	S
1067	H	MeO	A1	J11	S
1068	H	MeO	A1	J12	S
1069	H	MeO	A1	J13	S
1070	H	MeO	A1	J14	S
1071	H	MeO	A1	J15	S
1072	H	MeO	A1	J16	S
1073	H	MeO	A1	J17	S
1074	H	MeO	A1	J18	S
1075	H	MeO	A1	J19	S
1076	H	MeO	A1	J20	S
1077	H	MeO	A1	J21	S
1078	H	MeO	A1	J22	S
1079	H	MeO	A1	J23	S
1080	H	MeO	A1	J24	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1081	H	MeO	A1	J25	S
1082	H	MeO	A1	J26	S
1083	H	MeO	A1	J27	S
1084	H	MeO	A1	J28	S
1085	H	MeO	A1	J29	S
1086	H	MeO	A1	J30	S
1087	H	MeO	A1	J31	S
1088	H	MeO	A1	J32	S
1089	H	MeO	A2	J1	S
1090	H	MeO	A2	J2	S
1091	H	MeO	A2	J3	S
1092	H	MeO	A2	J4	S
1093	H	MeO	A2	J5	S
1094	H	MeO	A2	J6	S
1095	H	MeO	A2	J7	S
1096	H	MeO	A2	J8	S
1097	H	MeO	A2	J9	S
1098	H	MeO	A2	J10	S
1099	H	MeO	A2	J11	S
1100	H	MeO	A2	J12	S
1101	H	MeO	A2	J13	S
1102	H	MeO	A2	J14	S
1103	H	MeO	A2	J15	S
1104	H	MeO	A2	J16	S

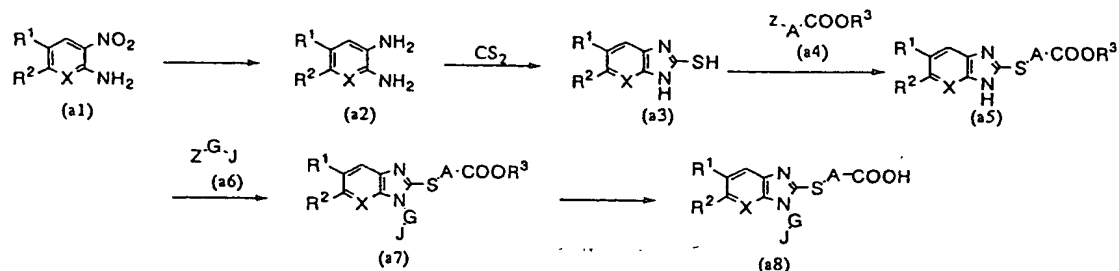
Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1105	H	MeO	A2	J17	S
1106	H	MeO	A2	J18	S
1107	H	MeO	A2	J19	S
1108	H	MeO	A2	J20	S
1109	H	MeO	A2	J21	S
1110	H	MeO	A2	J22	S
1111	H	MeO	A2	J23	S
1112	H	MeO	A2	J24	S
1113	H	MeO	A3	J25	S
1114	H	MeO	A2	J26	S
1115	H	MeO	A2	J27	S
1116	H	MeO	A2	J28	S
1117	H	MeO	A2	J29	S
1118	H	MeO	A2	J30	S
1119	H	MeO	A2	J31	S
1120	H	MeO	A2	J32	S
1121	H	MeO	A3	J1	S
1122	H	MeO	A3	J2	S
1123	H	MeO	A3	J3	S
1124	H	MeO	A3	J4	S
1125	H	MeO	A3	J5	S
1126	H	MeO	A3	J6	S
1127	H	MeO	A3	J7	S
1128	H	MeO	A3	J8	S

Compound No.	R <sub>1</sub>	R <sub>2</sub>	A	J	M
1129	H	MeO	A3	J9	S
1130	H	MeO	A3	J10	S
1131	H	MeO	A3	J11	S
1132	H	MeO	A3	J12	S
1133	H	MeO	A3	J13	S
1134	H	MeO	A3	J14	S
1135	H	MeO	A3	J15	S
1136	H	MeO	A3	J16	S
1137	H	MeO	A3	J17	S
1138	H	MeO	A3	J18	S
1139	H	MeO	A3	J19	S
1140	H	MeO	A3	J20	S
1141	H	MeO	A3	J21	S
1142	H	MeO	A3	J22	S
1143	H	MeO	A3	J23	S
1144	H	MeO	A3	J24	S
1145	H	MeO	A3	J25	S
1146	H	MeO	A3	J26	S
1147	H	MeO	A3	J27	S
1148	H	MeO	A3	J28	S
1149	H	MeO	A3	J29	S
1150	H	MeO	A3	J30	S
1151	H	MeO	A3	J31	S
1152	H	MeO	A3	J32	S

The benzimidazole derivative (1) of the present invention can be produced by synthesis method (A) or

synthesis method (B) shown below in the case E is COOR<sup>3</sup> and M is S:

Synthesis Method (A)



wherein, Z represents a halogen or ammonium group, and R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, A, G, J and X are as defined above.

Namely, an orthophenylenediamine compound (a2) is obtained by reducing the nitro group of a 2-nitroaniline derivative (a1). After reacting this with CS<sub>2</sub> and obtaining compound (a3), it is reacted with a halide ester derivative (a4) to obtain (a5) followed by further reacting with a halide derivative or ammonium salt (a6) to be able to obtain the compound (a7) of the present invention. In addition, benzimidazole derivative (a8), in which R<sup>3</sup> is a hydrogen atom, can be obtained by hydrolyzing this as necessary.

Reduction of the nitro group can be carried out, in accordance with the conditions of an ordinary catalytic reduction, by reacting with hydrogen gas at a temperature of room temperature to 100°C in the presence of a catalyst such as Pd-C under acidic, neutral or alkaline conditions. In addition, this can also be carried out by a method in which treatment is carried out using zinc or tin under acidic conditions, or by a method that uses zinc powder under neutral or alkaline conditions.

The reaction between orthophenylenediamine derivative (a2) and CS<sub>2</sub> can be carried out according to the method described in, for example, The Journal of Organic Chemistry (J. Org. Chem.), 1954, Vol. 19, pages 631-637 (pyridine solution) or in The Journal of Medical Chemistry (J. Med. Chem.), 1993, Vol. 36, pages 1175-1187

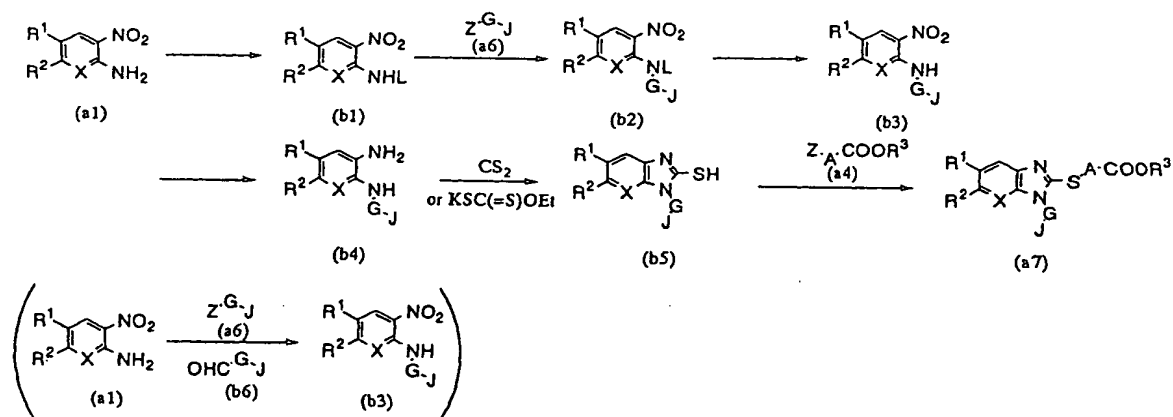
(ethanol solution).

The reaction between thiobenzimidazole compound (a3) and halide ester (a4) can be carried out by agitating at a temperature of 0°C-200°C in the presence of a base such as NaH, Et<sub>3</sub>N, NaOH or K<sub>2</sub>CO<sub>3</sub> in accordance with the conditions of an ordinary S-alkylation reaction.

The reaction between thiobenzimidazole compound (a5) and halide derivative or ammonium salt (a6) can be carried out by agitating at a temperature of 0°C-200°C in the presence of a base such as NaH, Et<sub>3</sub>N, NaOH, K<sub>2</sub>CO<sub>3</sub> or Cs<sub>2</sub>CO<sub>3</sub> in accordance with the conditions of an ordinary N-alkylation or N-acylation reaction.

A hydrolysis method using an alkali such as lithium hydroxide or an acid such as hydrochloric acid or trifluoroacetic acid is preferably used for the elimination reaction of the carboxy protecting group R<sup>3</sup>.

#### Synthesis Method (B)



Namely, (b1) is obtained by protecting the amino group of 2-nitroaniline derivative (a1) with a suitable protecting group L. This is then reacted with a halide derivative or ammonium salt (a6) to obtain (b2), and (b3) is obtained by removing protecting group L. An orthophenylenediamine derivative (b4) is obtained by reducing the nitro group of (b3). After reacting this with CS<sub>2</sub> or KSC(=S)Oet and obtaining compound (b5), this is reacted with a halide ester derivative (a4) to be able to obtain the benzimidazole derivative (a7) of the

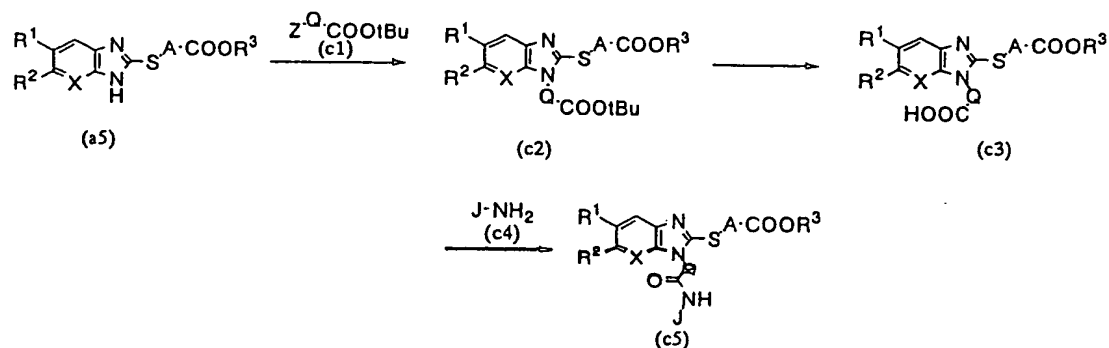


present invention. In addition, this may then be hydrolyzed as necessary to obtain a benzimidazole derivative of the present invention in which R<sup>3</sup> is a hydrogen atom.

5 Compound (b3) can also be obtained directly by reacting an unprotected halide derivative, ammonium salt (a6) or aldehyde derivative (b6) with the 2-nitroaniline derivative (a1). Examples of protecting group L include a trifluoroacetyl group, acetyl group, t-butoxycarbonyl group and benzyl group. The reaction between the 2-nitroaniline derivative (a1) and aldehyde derivative (b6) can be carried out by an ordinary reduction amination under temperature conditions of 0°C-200°C in a solvent such as ethanol, methanol or dichloromethane using a multiple hydrogen compound such as LiAlH<sub>4</sub>, NaBH<sub>4</sub>, NaBH<sub>3</sub>CN or NABH(OAc)<sub>3</sub> or a reducing agent such as diborane. In addition, the reaction between the orthophenylenediamine derivative (b4) and CS<sub>2</sub> can be carried out in the same manner as Synthesis Method (A), while the reaction with KSC(=S)Oet can be carried out according to the method described in, for example, Organic Synthesis (OS), 1963, Vol. 4, pages 569-570. Other reactions can be carried out in the same manner as Synthesis Method (A).

25 Benzimidazole derivative (1) of the present invention can be produced according to the following Synthesis Method (C) in the case E is COOR<sup>3</sup>, M is S and G is an amide bond:

#### Synthesis Method (C)



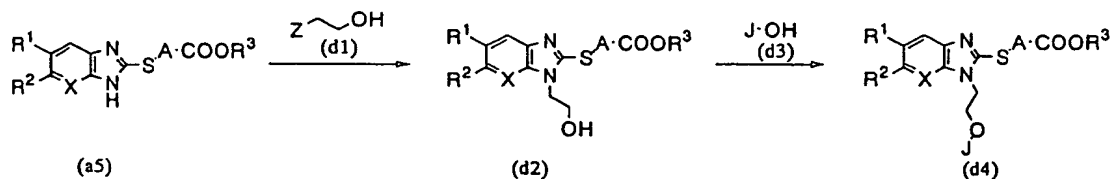
wherein, Q represents a methylene group, phenylene group, etc., Z represents a halogen, and R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, A, J and X are as defined above, provided that R<sup>3</sup> is an inactive protecting group such as a methyl group or ethyl group under acidic conditions.

Namely, compound (c2) is obtained by reacting thiobenzimidazole compound (a5) with tert-butylesterhalide derivative (c1). This is then hydrolyzed under acidic conditions to obtain (c3). This is then condensed with amine derivative (c4) to be able to obtain compound (c5) of the present invention. In addition, this may be hydrolyzed as necessary to obtain the benzimidazole derivative of the present invention in which R<sup>3</sup> is a hydrogen atom.

A typical method using a condensation agent is used for the conditions of condensation amidation. Examples of condensation agents include DCC, DIPC, EDC=WSCI, WSCI·HCl, BOP and DPPA, and these may be used alone or as a combination with HONSu, HOBT or HOObt. The reaction is carried out under temperature conditions of 0°C-200°C in a suitable solvent such as THF, chloroform or t-butanol. The other reactions can be carried out in the same manner as Synthesis Method (A).

The benzimidazole derivative (1) of the present invention can be produced according to the following Synthesis Method (D) in the case E is COOR<sup>3</sup>, M is S and G has an ether bond:

#### Synthesis Method (D)



wherein, Z represents a halogen and R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, A, J and X are as defined above.

Namely, compound (d2) is obtained by reacting, for example, halide alcohol derivative (d1) with

thiobenzimidazole compound (a5). This is then reacted with phenol derivative (d3) to be able to obtain compound (d4) of the present invention. In addition, this may be hydrolyzed as necessary to obtain a benzimidazole derivative in which R<sup>3</sup> is a hydrogen atom.

The etherification reaction is carried out by Mitsunobu reaction and similar reactions under temperature conditions of 0°C-200°C in a suitable solvent such as N-methylmorpholine or THF using a phosphine compound such as triphenylphosphine or tributylphosphine and an azo compound such as DEAD or TMAD. The other reactions can be carried out in the same manner as Synthesis Method (A).

Benzimidazole derivative (1) of the present invention can be produced according to the following Synthesis Method (E) in the case E is tetrazole-5-yl and M is S:

#### Synthesis Method (E)



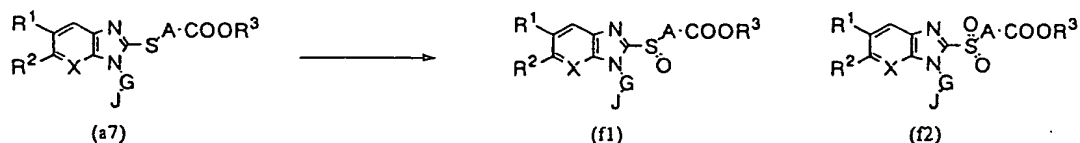
wherein, R<sup>1</sup>, R<sup>2</sup>, A, G, J and X are as defined above.

The nitrile form (e1) is converted to the tetrazole form (e2) by reacting with various azide compounds. Examples of azide compounds include trialkyl tin azide compounds such as trimethyl tin azide, hydrazoic acid and its ammonium salts. When using an organic tin azide compound, roughly 1-4 moles should be used relative to compound (e1). In addition, when using hydrazoic acid or its ammonium salts, 1-5 moles of sodium azide and ammonium chloride or tertiary amine such as triethylamine should be used relative to compound (e1). Each reaction is carried out at a temperature of 0°C-200°C by using a solvent such as toluene, benzene or DMF.

Benzimidazole derivative (1) of the present invention can be produced according to the following

Synthesis Method (F) in the case M is SO or SO<sub>2</sub>:

Synthesis Method (F)

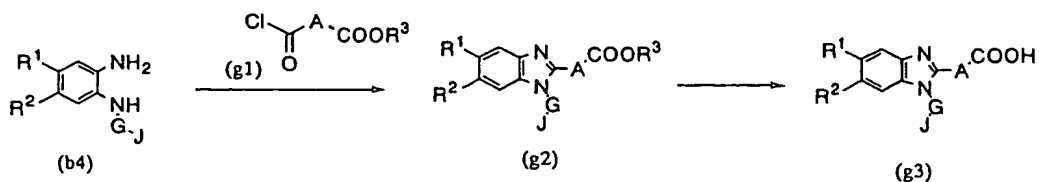


wherein, R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, A, G, J and X are as defined above.

Namely, sulfoxide derivative (a7) and/or sulfone derivative (f2) are obtained by reacting benzimidazole derivative (a7) with a peroxide compound in a suitable solvent. Examples of peroxide compounds used include perbenzoic acid, m-chloroperbenzoic acid, peracetic acid and hydrogen peroxide, while examples of solvents used include chloroform and dichloromethane. There are no particular restrictions on the usage ratio of compound (a7) and a peroxide compound, and although the ratio should be suitably selected over a wide range, it is generally preferable to use about 1.2 to 5 moles. Each reaction is normally carried out at 0-50°C, and preferably from 0°C to room temperature, and each reaction is typically completed in about 4-20 hours.

Benzimidazole derivative (1) of the present invention can be produced according to the following Synthesis Method (G) in the case M is a single bond:

Synthesis Method (G)



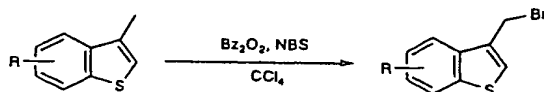
wherein, X, A, G, J and R<sup>3</sup> are as defined above.

Namely, benzimidazole derivative (g2) of the present invention can be obtained by reacting a known acid chloride derivative (g1) with a diamine compound (b4). In addition, hydrolyzing -COOR<sup>3</sup> of (g2) as necessary allows the obtaining of benzimidazole derivative (g3) in which R<sup>3</sup> is a hydrogen atom.

Furthermore, the cyclization reaction is described in the Journal of Medical Chemistry (J. Med. Chem.), 1993, Vol. 36, pages 1175-1187.

5 In addition, Z-G-J described in synthesis methods (A) through (F) can be synthesized by referring to a large number of publications.

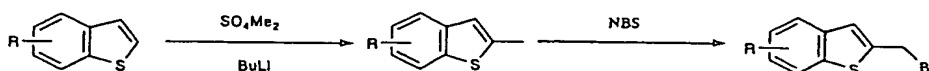
For example, a benzothiophene halide derivative can be synthesized by referring to the following literature and patent specification.



10

J. Chem. Soc. (1965), 774

J. Chem. Soc. Perkin Trans 1, (1972), 3011



JACS, 74, 664, (1951); Patent US4282227

These compounds can also be synthesized by referring to the following literature and patent specifications.

15 Namely, these compounds can be synthesized not only by the reactions described in the following literature, but also by combining typical reactions such as oxidation-reduction or OH halogenation.

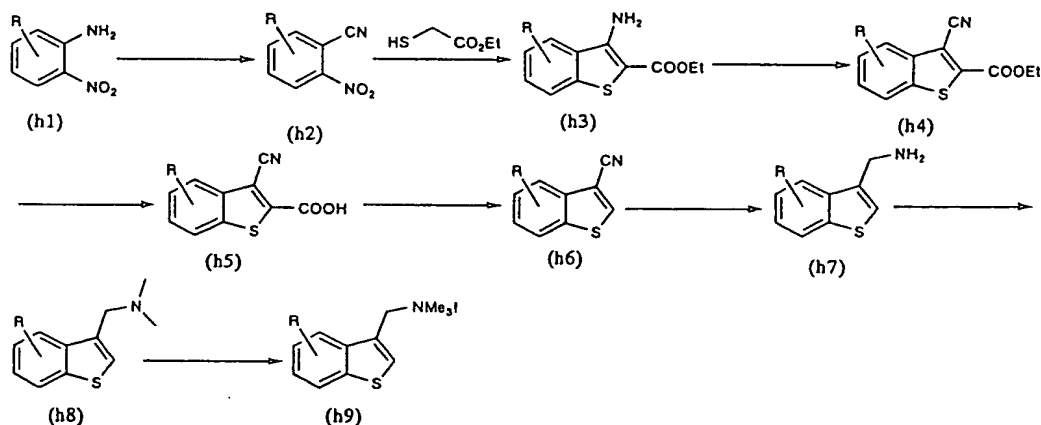
20 J Chem Soc, (1965), 774; Bull Chem Soc Jpn (1968), 41, 2215; Japanese Unexamined Patent Publication No. 10-298180; Sulfur Reports, (1999), Vol. 22, 1-47; J Chem Soc comm., (1988), 888; J. Heterocyclic Chem., 19, 859, (1982); Synthetic Communication, (1991), 21, 959; Tetrahedron Letters, (1992), Vol. 33, No. 49, 7499;  
25 Synthetic Communications, (1993), 23(6), 743; Japanese Unexamined Patent Publication No. 2000-239270; J. Med. Chem., (1985), 28, 1896; Arch Pharm, (1975), 308, 7, 513; Khim Gerotsikl Soedin, (1973), 8, 1026; Bull. Chem. Soc. Jpn., (1997), 70, 891; J. Chem. Soc. Perkin1, (1973),  
30 750; J. Chem. Soc. Chem. Comm., (1974), 849; J. Chem. Soc. Comm. (1972), 83

In particular, the hydroxymethyl form at position 3 of benzothiophene can be synthesized easily by referring to J. Chem. Soc. Chem. Comm., (1974), 849.

With respect to iodides, the Cl and Br forms can be obtained by halogen exchange with NaI and so forth.

In addition, the quaternary ammonium salt derivative of benzothiophene can be synthesized by reacting a suitable amine such as dimethylamine with the previously mentioned benzothiophene halide derivative. In addition, it may also be synthesized in the following manner:

#### Synthesis Method (H)



wherein, R represents one or more substituents in the above-mentioned J, the number of substituents is optional, and the substituents may be independent substituents.

Namely, cyclic benzothiophene derivative (h3) is obtained by converting the amino group of 2-nitroaniline derivative (h1) to a cyano form (h2) and reacting with ethyl 2-mercaptoacetate. Moreover, carboxylic acid (h5) is obtained by cyanating the amino group to a cyano form (h4) followed by ester hydrolysis. The carboxylic acid is then decarboxylated to obtain (h6). Continuing, the cyano group is reduced to convert to an amino form (h7) followed by N-dimethylation to obtain (h8) and then followed by N-methylation to be able to obtain quaternary salt (h9).

Cyanation of the amino group of 2-nitroaniline

derivative (h1) by converting the amino group to  
diazonium using, for example, hydrochloric acid or sodium  
sulfite, and then further reacting with copper (I)  
chloride and potassium cyanide to convert to the cyano  
form.

5  
Reaction from cyano form (h2) to benzothiophene  
derivative (h3) can be carried out to obtain cyclic  
benzothiophene derivative (h3) by heating with ethyl 2-  
mercaptoacetate in a suitable solvent such as DMF in the  
10 presence of a suitable basic reagent by referring to the  
method described in, for example, Synthetic  
Communications, 23(6), 743-748 (1993); or Farmaco, Ed.  
Sci., 43, 1165 (1988).

With respect to the cyanation of (h3), (h3) can be  
15 converted to the cyano form (h4) by reacting copper  
cyanide and t-butyl sulfite in a suitable solvent such as  
DMSO under suitable temperature conditions.

Ester hydrolysis can be carried out by routinely  
used methods. For example, carboxylic acid (h5) can be  
20 obtained by ester hydrolysis in a suitable solvent such  
as THF-MeOH in the presence of a suitable basic reagent  
such as sodium hydroxide.

The carboxylic acid decarboxylation reaction can be  
carried out by heating in a suitable solvent such as  
25 quinoline solvent in the presence of a copper catalyst.

Reduction of the cyano group to an amino group can  
be carried out to obtain the amino form by, for example,  
reducing in a suitable solvent such as Et<sub>2</sub>O-THF under  
suitable temperature conditions using a suitable reducing  
30 agent such as lithium aluminum hydride.

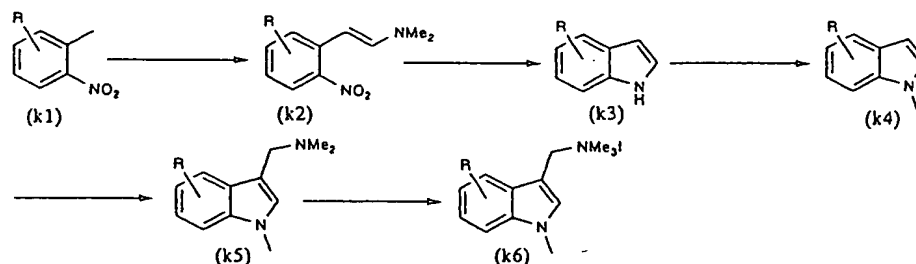
Methylation of the amino group can be carried out by  
heating in, for example, formic acid or aqueous formalin  
solution.

Conversion of the amino group to a quaternary salt  
35 can be carried out by, for example, reacting with methyl  
iodide in ethanol solvent.

Indole quaternary amine salt derivative can be

synthesized according to, for example, the following method:

Synthesis Method (K)



wherein, R represents one or more substituents in the above-mentioned J, the number of substituents is optional, and the substituents may be independent substituents.

Namely, nitro form (k1) is converted to an enamine (k2) by enamination followed by converting to the indole form (k3) by indole cyclization according to the method of Reissert. Moreover, the 3<sup>rd</sup> position dimethylaminomethyl form (k5) is obtained according to the Mannich reaction following N-dimethylation and this is followed by N-methylation to be able to obtain the quaternary amine salt (k6).

The enamination reaction can be carried out by, for example, heating the O-nitrotoluene derivative (k1) with N,N-dimethylformamide dimethylacetal and pyrrolidine in a suitable solvent such as DMF.

The indole cyclization reaction can be carried out by reacting at room temperature using hydrogen gas in the presence of Raney nickel in a suitable solvent such as toluene.

N-methylation can be carried out by, for example, heating in DMF solvent using t-butoxypotassium or dimethyl oxalate.

3<sup>rd</sup> position dimethylaminomethylation can be carried out by, for example, using the Mannich reaction and reacting at room temperature in dioxane-acetic acid solvent using aqueous formalin solution or aqueous



dimethylamine solution.

In addition, the indole derivative can be synthesized by referring to the literature of Heterocycles, Vol. 22, No. 1, 195, (1984).

5        Moreover, benzothiophene, indole and other heterocyclic halides and quaternary salts can be synthesized by referring to other references in the literature such as Heterocyclic Compound Chemistry, (Kondansha Scientific, H. Yamanaka, ed.).

10        Benzimidazole derivative of the present invention can also be converted to a medically acceptable, non-toxic cation salt as necessary. Examples of such salts include alkali metal ions such as  $\text{Na}^+$  and  $\text{K}^+$ , alkaline earth metal ions such as  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  and metal ions such as  $\text{Al}^{3+}$  and  $\text{Zn}^{2+}$ , as well as organic bases such as ammonia, triethylamine, ethylenediamine, propanediamine, pyrrolidine, piperidine, piperazine, pyridine, lysine, choline, ethanolamine, N,N-dimethylethanolamine, 4-hydroxypiperidine, glucosamine and N-methylglucamine. In  
15        particular,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ , lysine, choline, N,N-dimethylethanolamine and N-methylglucamine are preferable.

20        Benzimidazole derivative of the present invention strongly inhibits human chymase activity. More specifically,  $\text{IC}_{50}$  is 1000 nM or less, preferably 0.01 nM or more to less than 1000 nM, and more preferably 0.05 nM or more to less than 500 nM. The benzimidazole derivative of the present invention having such superior human chymase inhibitory activity can be used as a  
25        preventive agent and/or therapeutic agent clinically applicable to various diseases.

30        Benzimidazole derivative of the present invention can be administered orally or non-orally as a pharmaceutical composition together with a  
35        pharmaceutically allowed carrier by forming said pharmaceutical composition into various drug forms. Examples of non-oral administration include intravenous,

subcutaneous, intramuscular, transcutaneous, rectal, nasal and intraocular administration.

5        Examples of drug forms of said pharmaceutical composition include tablets, pills, granules, powders, liquids, suspensions, syrups and capsules in the case of oral preparations.

10       Here, for the method of forming tablets, tablets can be formed by an ordinary method using a pharmaceutically acceptable carrier such as a vehicle, binder or disintegration agent. Pills, granules and powders can be formed by an ordinary method using a vehicle and so forth in the same manner as tablets. Liquids, suspensions and syrups can be formed according to an ordinary method using glycerin esters, alcohols, water or vegetable oil.

15       Capsules can be formed by filling granules, powders or liquids and so forth into capsules made of gelatin and so forth.

20       Non-oral preparations can be administered in the form of an injection preparation in the case of administration by intravenous, subcutaneous or intramuscular administration. Examples of injection preparations include the case in which a benzoic acid derivative is dissolved in a water-soluble liquid agent such as physiological saline, or the case in which it is dissolved in a non-aqueous liquid agent composed of an organic ester such as vegetable oil.

25       In the case of percutaneous administration, a drug form such as an ointment or cream can be used. Ointments can be formed by mixing a benzoic acid derivative with an oil or Vaseline and so forth, while creams can be formed by mixing a benzoic acid derivative with an emulsifier.

30       In the case of rectal administration, administration can be performed in the form of a suppository using gelatin soft capsules and so forth.

35       In the case of nasal administration, a preparation can be used that is composed of a liquid or powder composition. Examples of bases of liquid preparations

that are used include water, saline, phosphate buffer and acetate buffer, and may also contain surfactant, antioxidant, stabilizer, preservative or thickener. Examples of bases of powdered preparations include

5 moisture absorbing bases such as water-soluble polyacrylates, cellulose lower alkyl ethers, polyethylene glycol polyvinyl pyrrolidone, amylose and plurane, or water-insoluble bases such as celluloses, starches, proteins, rubbers and cross-linked vinyl polymers,

10 although water-soluble bases are preferable. In addition, these may also be used as a mixture. Moreover, antioxidant, colorant, preservative, antiseptic or polysaprobic agent may be added to powdered preparations. Said liquid preparations and powdered preparations can be

15 administered using a sprayer and so forth.

In the case of intraocular administration, an aqueous or non-aqueous eye wash can be used. Aqueous eye washes can use sterile purified water or physiological saline for the solvent. In the case of using only

20 sterile purified water for the solvent, it can be used in the form of an aqueous suspended eye wash by adding surfactant, polymer thickener and so forth. In addition, it can also be used in the form of a soluble eye wash by adding a solubilization agent such as non-ionic

25 surfactant. Non-aqueous eye washes can use a non-aqueous solvent for injection for the solvent, and can also be used in the form of a non-aqueous suspended eye wash.

Examples of drug forms used in the case of administration to the eye other than as an eye wash

30 include ophthalmic ointments, applied liquids, sprays and inserts.

In addition, in the case of inhaling through the nose or mouth, benzimidazole derivative of the present invention is inhaled using, for example, an aerosol spray

35 for inhalation by combining with a typically used pharmaceutical vehicle in the form of a solution or suspension. In addition, benzimidazole derivative of the

present invention can be administered in the form of a dry powder using an inhaler in direct contact with the lungs.

5 Pharmaceutically allowed carriers such as isotonic agents, preservatives, antiseptics, moisturizers, buffers, emulsifiers, dispersants and stabilizers can be added to these various preparations as necessary.

10 In addition, these various preparations can be sterilized by performing treatment such as blending with a disinfectant, filtering using a bacteria-trapping filter, heating or irradiating as necessary. Alternatively, a sterile solid preparation can be produced and used after dissolving or suspending in a suitable sterile liquid immediately prior to use.

15 Although the dosage of benzimidazole derivative of the present invention varies according to the type of disease, administration route and symptoms, age, sex and body weight, etc. of the patient, it is typically 1-500 mg/day/person, and preferably 10-300 mg/day/person, in 20 the case of oral administration, and 0.1-100 mg/day/person, and preferably 0.3-30 mg/day/person, in non-oral administration such as intravenous, subcutaneous, intramuscular, percutaneous, rectal, nasal, intraocular or inhalation administration.

25 In addition, in the case of using benzimidazole derivative of the present invention as a preventive agent, it can be administered in accordance with previously known methods according to each symptom.

30 Examples of target diseases of the preventive agent and/or therapeutic agent of the present invention include respiratory diseases such as bronchial asthma, inflammatory and allergic diseases such as allergic rhinitis, atopic dermatitis and urticaria, cardiovascular diseases such as sclerosing vascular lesions, 35 vasoconstriction, peripheral circulatory disorders, renal insufficiency and cardiac insufficiency, and bone and cartilage metabolic diseases such as rheumatoid arthritis

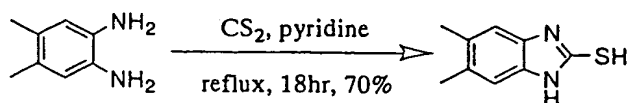
and osteoarthritis.

### Examples

5 The following provides a detailed explanation of the present invention according to its production examples, examples and test examples. However, the scope of the present invention is not restricted in any sense by these examples.

[Reference Example 1]

#### Production of 5,6-dimethylbenzimidazole-2-thiol

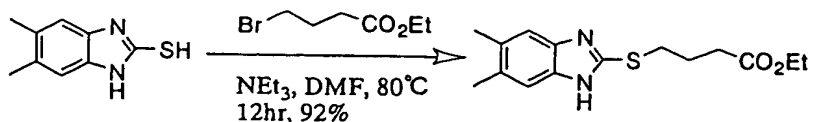


10 40 ml (0.66 mmol) of carbon disulfide were added to a pyridine solution (40 ml) of 4.5 g (33 mmol) of 5,6-dimethylorthophenylenediamine. After stirring the resulting solution for 18 hours while refluxing under heating, water was added followed by extraction with  
15 ethyl acetate. After drying the ethyl acetate phase with anhydrous magnesium sulfate, it was concentrated under reduced pressure and dried for 6 hours at 80°C under reduced pressure to obtain 4.1 g of the target compound (yield: 70%).

20 <sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 12.30 (br, 1H), 6.91 (s, 2H), 2.21 (s, 6H)

[Reference Example 2]

#### Production of 4-(5,6-dimethylbenzimidazole-2-ylthio)butanoate ethyl ester



25 35 μl (0.25 mmol) of triethylamine and 36 μl (0.25 mmol) of 4-bromobutanoate ethyl ester were added to 36 mg (0.20 mmol) of the obtained 5,6-dimethylbenzimidazole-2-thiol. After stirring the resulting solution for 12 hours at 80°C, water was added followed by extraction

with diethyl ether. After drying the diethyl ether phase with anhydrous magnesium sulfate, it was concentrated and residue was purified by silica gel column chromatography (hexane:ethyl acetate = 4:1) to obtain 54 mg of the target compound (yield: 92%). Confirmation of the compound was carried out by identifying it from the molecular weight using LC-MS.

Calculated value M = 292.12, Measured value (M+H)<sup>+</sup> = 293.40

[Reference Example 3]

The following compounds were synthesized according to the same method as Reference Example 2. Confirmation of the compounds was carried out by identifying them from the molecular weight using LC-MS.

4-(benzimidazole-2-ylthio)butanoate ethyl ester

Calculated value M = 264.09, Measured value (M+H)<sup>+</sup> = 293.40

4-(5,6-difluorobenzimidazole-2-ylthio)butanoate ethyl ester

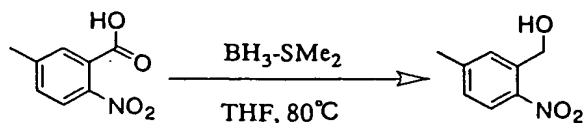
Calculated value M = 300.07, Measured value (M+H)<sup>+</sup> = 301.3

[Reference Example 4]

Production of 3-bromomethyl-5-methylbenzo[b]thiophene

Step 1

Production of 3-hydroxymethyl-p-nitrotoluene

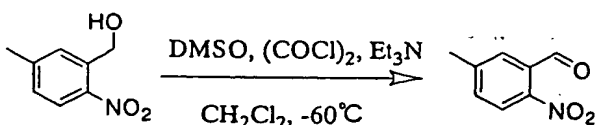


5.02 g (27.7 mmol) of 5-methyl-2-nitrobenzoic acid were dissolved in 20 ml of THF followed by dropping in 11.1 ml of 10.2 M borane dimethylsulfide complex and heating at 80°C. After 1.5 hours, 30 ml of 1 M hydrochloric acid were dropped into this reaction system while cooling with ice and stirring. The system was then concentrated under reduced pressure to obtain 100 ml of

the aqueous phase followed by extraction with ethyl acetate (100 ml x 2). After washing the ethyl acetate phase with saturated brine, the organic phase was dried with magnesium sulfate followed by concentration under reduced pressure and drying to obtain 3.91 g of the target compound (yield: 85%).

Step 2

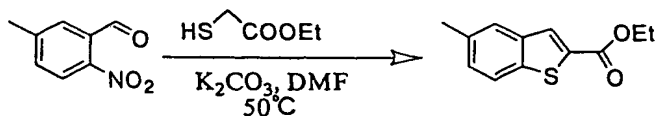
Production of 3-formyl-p-nitrotoluene



5.5 ml (63.2 mmol) of oxalyl chloride were added to 50 ml of dichloromethane and cooled to  $-60^{\circ}\text{C}$ . After 20 minutes, 9.13 ml (138.6 mmol) of DMSO were added and stirred at  $-60^{\circ}\text{C}$  followed 15 minutes later by the addition of 3.91 g (23.3 mmol) of the 3-hydroxymethyl-p-nitrotoluene obtained in Step 1 at  $-60^{\circ}\text{C}$  and stirring. After 30 minutes, 45 ml of triethylamine were dropped in at  $-60^{\circ}\text{C}$  and then returned to room temperature. After concentrating under reduced pressure, 0.1 M hydrochloric acid was added to the residue followed by extraction with ethyl acetate (150 ml x 2). The organic phase was then dried with magnesium sulfate and concentrated under reduced pressure to obtain 5.02 g of the target compound (crude yield: 130%).

Step 3

Production of 2-carboxyethyl-5-methylbenzo[b]thiophene



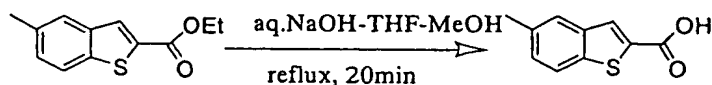
5.02 g (63.2 mmol) of the 3-formyl-p-nitrotoluene obtained in Step 2 were dissolved in 50 ml of DMF followed by the addition of 3.06 ml (28.1 mmol) of ethyl mercaptoacetate and 4.85 g (35.1 mmol) of potassium carbonate and stirring at  $50^{\circ}\text{C}$ . After 9.5 hours, the

temperature was raised to 80°C followed by additional heating for 100 minutes. Following completion of the reaction, 250 ml of water were added to the reaction solution followed by extraction with ethyl acetate (100 ml x 3) and drying with magnesium sulfate. After concentrating the solvent under reduced pressure, the residue was purified by silica gel column chromatography (hexane:ethyl acetate = 8:1) followed by additionally purifying by silica gel column chromatography (hexane:ethyl acetate = 10:1) to obtain 2.48 g (11.2 mmol) of the target compound (yield: 48%).

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 7.98 (s,1H), 7.73 (d,1H,J=8.28 Hz), 7.65 (s,1H), 7.27 (d,1H,J=8.32 Hz), 4.39 (q,2H), 2.47 (s,3H), 1.41 (s,3H)

#### Step 4

#### Production of 2-carboxy-5-methylbenzo[b]thiophene

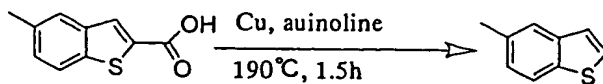


30 ml of a solution of methanol, THF and 2 M aqueous sodium hydroxide solution (1:1:1) were added to 2.17 g (9.87 mmol) of the 2-carboxyethyl-5-methylbenzo[b]thiophene obtained in Step 3 and refluxed. After 20 minutes, the solution was neutralized with acid followed by concentration under reduced pressure and recovery of the precipitate. This was then washed with 50 ml of water and dried to obtain 2.03 g (10.5 mmol) of the target compound (crude yield: 107%).

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 7.94 (s,1H), 7.74 (d,1H,J=8.56 Hz), 7.69 (s,1H), 7.27 (d,1H,J=8.30 Hz), 2.47 (s,3H)

#### Step 5

#### Production of 5-methylbenzo[b]thiophene



2.03 g (9.87 mmol) of the 2-carboxy-5-methylbenzo[b]thiophene obtained in Step 4 were dissolved

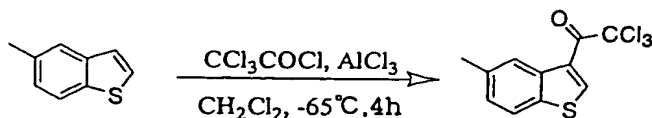


in 10 ml of quinoline followed by the addition of 799.2 mg of copper powder and heating to 190°C. After 100 minutes, the solution was cooled followed by the addition of 40 ml of 0.5 M hydrochloric acid and extraction with ethyl acetate (40 ml x 2). The organic phase was washed with 40 ml of water and then dried with magnesium sulfate. After concentrating the solvent under reduced pressure, it was purified by silica gel column chromatography (hexane:ethyl acetate = 20:1) to obtain 1.41 g (9.51 mmol) of the target compound (yield of the two steps from Step 4: 96%).

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.76 (d, 1H, J=8.24 Hz), 7.62 (s, 1H), 7.40 (d, 1H, J=5.44 Hz), 7.24 (m, 1H), 7.17 (d, 1H, J=8.24 Hz), 2.47 (s, 3H)

#### Step 6

#### Production of 3-chloromethylcarbonyl-5-methylbenzo[b]thiophene



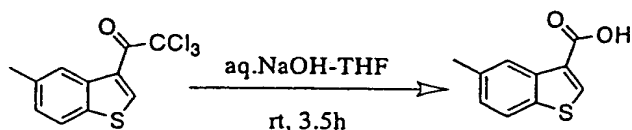
10 ml of dichloromethane were added to 1.33 g (9.97 mmol) of aluminum trichloride followed by cooling to -65°C with dry ice and acetone. After 10 minutes, 1.12 ml (10.0 mmol) of trichloroacetylchloride were dropped in. After an additional 20 minutes, 10 ml of dichloromethane solution containing 1.41 g (9.51 mmol) of the 5-methylbenzo[b]thiophene obtained in Step 5 were dropped in and then stirred at about -65°C. After 1 hour and 40 minutes, the temperature was raised to -40°C. After an additional 1 hour and 10 minutes, the temperature was raised to 0°C. After another 1 hour and 40 minutes, 10 ml of 1 M hydrochloric acid were added and stirred. After adding 20 ml of water to the reaction system, removing the dichloromethane phase by a liquid separation procedure and then additionally extracting the aqueous phase with ethyl acetate, the aqueous phase was combined with the dichloromethane phase and then

concentrated under reduced pressure. 3.2 g of the resulting residue were purified by silica gel column chromatography (silica gel: 120 g, hexane) to obtain 686.7 mg (2.34 mmol) of the target compound (yield: 24%).

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 8.89 (s,1H), 8.51 (s,1H), 7.78 (d,1H,J=8.28 Hz), 7.30 (d,1H,J=8.32 Hz), 2.53 (s,3H)

Step 7

Production of 3-carboxy-5-methylbenzo[b]thiophene

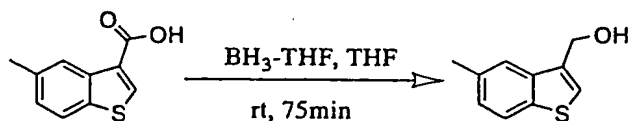


686.7 mg (2.34 mmol) of the 3-chloromethylcarbonyl-5-methylbenzo[b]thiophene obtained in Step 6 were dissolved in 2.0 ml of THF and 3.0 ml of MeOH followed by the addition of 2 ml of 2 M aqueous sodium hydroxide solution and stirring at room temperature. After 2 hours and 45 minutes, 5 ml of 2 M aqueous sodium hydroxide solution were added followed by heating to 60°C. After cooling 30 minutes later and adding 10 ml of 2 M hydrochloric acid and 30 ml of water, the solution was extracted with ethyl acetate followed by concentration under reduced pressure and drying to obtain 438.9 mg (2.28 mmol) of the target compound (yield: 97%).

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 8.44 (s,1H), 8.36 (s,1H), 7.74 (d,1H,J=8.04 Hz), 7.22 (d,1H,J=8.28 Hz), 2.50 (s,3H)

Step 8

Production of 3-hydroxymethyl-5-methylbenzo[b]thiophene



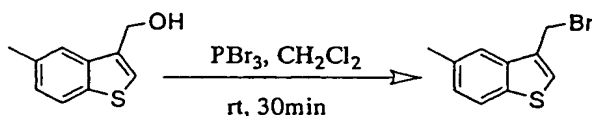
438.9 mg (2.28 mmol) of the 3-carboxy-5-methylbenzo[b]thiophene obtained in Step 7 were dissolved in 5 ml of THF followed by the addition of BH<sub>3</sub>·THF

complex solution and stirring at room temperature. After 1 hour and 15 minutes, 4 ml of 2 M hydrochloric acid were added and stirred followed by the addition of 50 ml of ethyl acetate. The organic phase was washed with 30 ml of water and dried with magnesium sulfate followed by concentration under reduced pressure. The resulting residue was purified with Biotage (hexane:ethyl acetate = 4:1) to obtain 347.6 mg (1.95 mmol) of the target compound (yield: 86%).

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 7.74 (d, 1H, J=8.04 Hz), 7.65 (s, 1H), 7.34 (s, 1H), 7.19 (d, 1H, J=8.28 Hz), 4.89 (s, 2H), 2.48 (s, 3H)

#### Step 9

#### Production of 3-bromomethyl-5-methylbenzo[b]thiophene



326 mg (1.83 mmol) of the 3-hydroxymethyl-5-methylbenzo[b]thiophene obtained in Step 8 were dissolved in 10 ml of dichloromethane followed by the addition of 0.262 ml of phosphorous tribromide and stirring at room temperature. After 30 minutes, 30 ml of water were added followed by stirring for 10 minutes and extracting with dichloromethane (30 ml x 2). The organic phase was then concentrated under reduced pressure and dried to obtain 397.5 mg (1.65 mmol) of the target compound (yield: 90%).

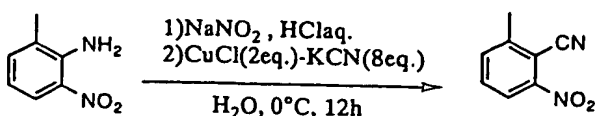
<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.74-7.67 (m, 2H), 7.46 (s, 1H), 7.22 (d, 1H, J=8.24 Hz), 4.74 (s, 2H), 2.51 (s, 3H)

#### [Reference Example 5]

#### Production of ((4-methylbenzo[b]thiophene-3-yl)methyl)trimethylammonium iodide

#### Step 1

#### Production of 2-cyano-3-nitrotoluene



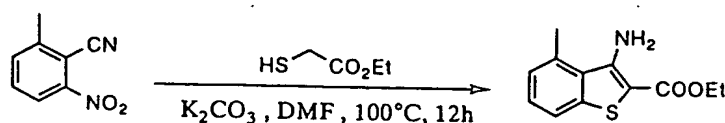
76.07 g (500 mmol) of 2-amino-3-nitrotoluene were added to 100 g (990 mmol) of 36% hydrochloric acid and 500 g of ice followed by stirring vigorously at 0°C. 80 ml of an aqueous solution containing 37.95 g (550 mmol) of sodium nitrite was then slowly dropped in while holding the temperature to 0-5°C. Following completion of dropping, 100 ml of toluene were added followed by stirring for 30 minutes at 0°C. The reaction solution was placed in an ice-NaCl bath followed by slowly adding sodium bicarbonate while stirring vigorously to neutralize the pH to about 6 (diazonium salt solution (1)).

An aqueous solution (550 ml) containing 260.5 g (4000 mmol) of potassium cyanide was slowly added at 0°C to an aqueous solution (650 ml) containing 99.0 g (1000 mmol) of copper (I) chloride followed by stirring for 90 minutes and then adding 200 ml of ethyl acetate. The diazonium salt solution (1) prepared above was then dropped into this solution over the course of 30 minutes while holding the temperature to 0-5°C. The solution was then stirred for 12 hours in an ice bath and then warmed to room temperature. After extracting the reaction solution with ethyl acetate and washing the organic phase with water, it was dried with magnesium sulfate followed by concentrating the solvent under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 20:1→10:1→7:1→5:1→3:1) to obtain 58.63 g (362 mmol) of the target compound (yield: 72%).

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.68 (2H,m), 8.13 (1H,m), 2.715 (3H,s)

#### Step 2

#### Production of 3-amino-2-ethoxycarbonyl-4-methylbenzo[b]thiophene

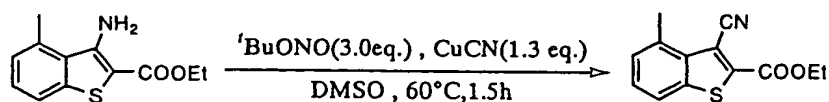


A DMF solution (250 ml) containing 58.63 g (362 mmol) of the 2-cyano-3-nitrotoluene obtained in Step 1, 47.5 g (395 mmol) of ethyl 2-mercaptoacetate and 57.5 g (416 mmol) of potassium carbonate was stirred for 12 hours at 100°C. The reaction solution was then concentrated, as is, under reduced pressure to remove the DMF to a certain degree. Water was added to dissolve inorganic substances followed by extraction with ethyl acetate. After washing the organic phase with water, it was dried with magnesium sulfate followed by concentration of the solvent under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 10:1) to obtain 62.86 g (267 mmol) of the target compound (yield: 74%).

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.54 (d, 1H), 7.29 (t, 1H), 7.03 (d, 1H), 6.28 (s, 2H), 4.35 (q, 2H), 2.82 (s, 3H), 1.39 (t, 3H)

### Step 3

#### Production of 3-cyano-2-ethoxycarbonyl-4-methylbenzo[b]thiophene



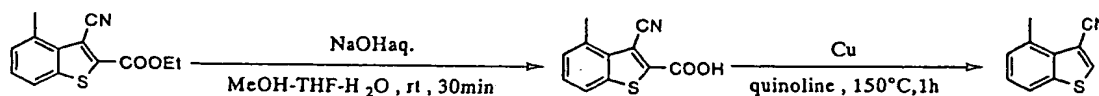
After replacing the reaction system with nitrogen, 82.0 g (795 mmol) of t-butyl nitrite and 30.9 g (345 mmol) of copper cyanide were added to 250 ml of DMSO and dissolved by stirring for 30 minutes at 55°C. Moreover, a DMSO solution (100 ml) containing 62.2 g (265 mmol) of the 3-amino-2-ethoxycarbonyl-4-methylbenzo[b]thiophene obtained in Step 2 was slowly dropped in over the course of 2 hours while holding the temperature at 55°C. After warming the reaction solution to 60°C and stirring for 140 minutes, it was cooled to 0°C followed by slowly

adding water and stirring for 1 hour at 0°C. The reaction solution was then filtered with Celite to remove impurities, and after extracting with dichloromethane and washing the organic phase with water, it was dried with magnesium sulfate followed by concentrating the solvent under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 20:1→15:1→ 10:1) to obtain 15.59 g (63.6 mmol) of the target compound (yield: 24%).

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.73 (d,1H), 7.44 (t,1H), 7.30 (d,1H), 4.50 (q,2H), 2.95 (s,3H), 1.47 (t,3H)

#### Step 4

#### Production of 3-cyano-4-methylbenzo[b]thiophene



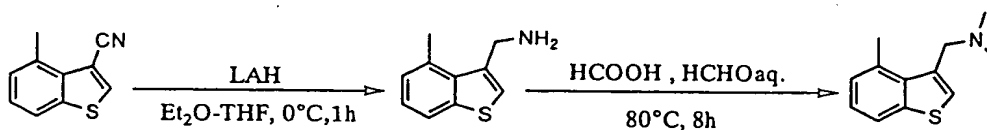
15.59 g (63.6 mmol) of the 3-cyano-2-ethoxycarbonyl-4-methylbenzo[b]thiophene obtained in Step 3 were dissolved in a mixture of methanol (150 ml), THF (150 ml) and water (150 ml) followed by the addition of 30 ml of 5 M aqueous sodium hydroxide solution and stirring for 2 hours at room temperature. After concentrating the solvent under reduced pressure, the pH was lowered to 4 by addition of 1 M hydrochloric acid and, after extracting with ethyl acetate and washing the organic phase with water, it was dried with magnesium sulfate. The solvent was then concentrated under reduced pressure to obtain 3-cyano-2-carboxy-4-methylbenzo[b]thiophene. This and 1.27 g (20 mmol) of copper powder were added to 18 ml of quinoline followed by stirring for 2 hours at 150°C. After cooling the reaction solution, it was filtered with Celite and the pH of the filtrate was lowered to 3 by addition of hydrochloric acid to transfer the quinoline as the solvent to the aqueous phase followed by extraction with ethyl acetate. After washing the organic phase with water, it was dried with magnesium

sulfate and the solvent was concentrated under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 20:1) to obtain 9.10 g (52.6 mmol) of the target compound (yield of the two steps: 83%).

$^1\text{H-NMR}$  (270 MHz,  $\text{CDCl}_3$ ) (ppm): 8.15 (s, 1H), 7.74 (d, 1H), 7.36 (t, 1H), 7.25 (d, 1H), 2.91 (s, 3H)

Step 5

Production of 3-((N,N-dimethylamino)methyl)-4-methylbenzo[b]thiophene



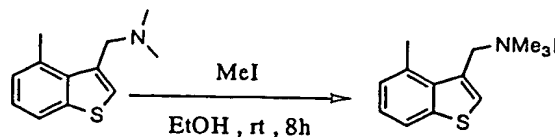
After dropping a diethyl ether (20 ml) and THF (20 ml) solution containing 9.10 g (52.6 mmol) of the 3-cyano-4-methylbenzo[b]thiophene obtained in Step 4 into 50 ml of a diethyl ether suspension of 2.0 g (53 mmol) of lithium aluminum hydride over the course of 15 minutes at 0°C, the solution was stirred for 30 minutes at room temperature. Following completion of the reaction, excess LAH in the reaction solution was treated with hydrochloric acid followed by the addition of aqueous sodium hydroxide solution to make alkaline. After saturating the aqueous phase with potassium carbonate, extracting with dichloromethane and washing the organic phase with water, it was dried with magnesium sulfate. The solvent was then concentrated under reduced pressure to obtain 3-aminomethyl-4-methylbenzo[b]thiophene. 11.5 (250 mmol) of formic acid and 10.0 g (123 mmol) of 37% aqueous formaldehyde solution were sequentially added to this followed by stirring for 5 hours at 80°C. Following completion of the reaction, after adding aqueous hydrochloric acid solution to the reaction solution, it was concentrated under reduced pressure to remove the formic acid and formaldehyde. Aqueous sodium hydroxide solution was then added to make the solution alkaline

followed by extraction with dichloromethane. After washing the organic phase with water, it was dried with magnesium sulfate and the solvent was concentrated under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 10:1) to obtain 2.61 g (12.8 mmol) of the target compound (yield of the two steps: 24%). Confirmation of the compound was carried out by identifying from <sup>1</sup>H-NMR.

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.66 (s,1H), 7.26-7.09 (m,3H), 3.65 (s,2H), 2.85 (s,3H), 2.27 (s,6H)

Step 6

Production of ((4-methylbenzo[b]thiophene-3-yl)methyl)trimethylammonium iodide



3.69 g (26 mmol) of methyl iodide were added to 20 ml of an ethanol solution containing 2.61 g (12.8 mmol) of the 3-((N,N-dimethylamino)methyl)-4-methylbenzo[b]thiophene obtained in Step 5 followed by stirring for 18 hours at room temperature. As this results in a white suspension, after filtering out the excess methyl iodide and solvent, it was washed with ethanol (10 ml x 2) and diethyl ether (10 ml x 3) to obtain 3.08 g (8.88 mmol) of the target compound in the form of a white solid (yield: 69%).

<sup>1</sup>H-NMR (270 MHz, DMSO)(ppm): 8.19 (s,1H), 7.93 (d,1H), 7.36-7.25 (m,2H), 4.91 (s,2H), 3.05 (s,9H), 2.77 (s,3H)

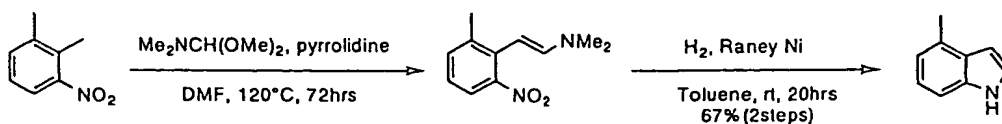
[Reference Example 6]

Production ((1,4-dimethylindole-3-yl)methyl)methylammonium iodide

Step 1

Production of 4-methylindole



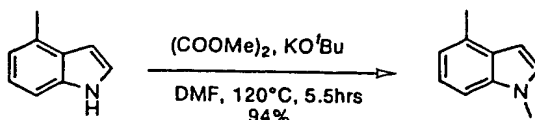


30.5 g (256 mmol) of N,N-dimethylformamidedimethylacetal and 10.9 g (153 mmol) of pyrrolidine were added to 150 ml of an N,N-dimethylformamide solution containing 19.4 g (128 mmol) of 2,3-dimethylnitrobenzene. After stirring the resulting solution for 72 hours at  $120^\circ\text{C}$ , it was concentrated as is. 100 ml of toluene were added to the resulting brown oily substance followed by the addition of 11 g of Raney nickel (50%, aqueous slurry, pH > 9) and stirring. The inside of the reaction vessel was replaced with hydrogen gas followed by stirring for 20 hours at room temperature in a hydrogen gas atmosphere. After filtering the reaction solution with Celite, the filtrate was concentrated to obtain 30 g of a black solution. This was then purified by silica gel column chromatography (hexane:ethyl acetate = 10:1) to obtain 11.33 g (86 mmol) of the target compound (yield of the two steps: 67%). Confirmation of the compound was carried out by identifying using  $^1\text{H-NMR}$ .

$^1\text{H-NMR}$  (270 MHz,  $\text{CDCl}_3$ ) (ppm): 7.28-7.07 (m, 3H), 6.93 (m, 1H), 6.57 (m, 1H), 2.57 (s, 3H)

#### Step 2

#### Production of 1,4-dimethylindole



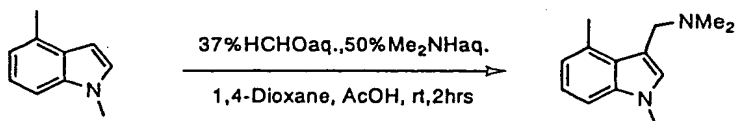
12.7 g (134 mmol) of t-butoxypotassium and 80 ml of N,N-dimethylformamide were added to a pre-dried reaction vessel. 8.9 g (67.9 mmol) of the 4-methylindole obtained in Step 1 were added followed by stirring for 35 minutes at room temperature. 15.8 g (134 mmol) of dimethyl oxalate were added to this followed by stirring for 5 hours and 30 minutes at  $120^\circ\text{C}$ . After concentrating under

reduced pressure, 200 ml of water were added followed by treatment with 1 M hydrochloric acid to make acidic (pH = 3) followed by extraction with ethyl acetate (200 ml x 2) and drying with anhydrous magnesium sulfate. After  
5 distilling off the solvent under reduced pressure, it was purified by silica gel column chromatography (hexane:ethyl acetate = 5:1) to obtain 9.24 g (53 mmol) of the target compound (yield: 94%). Confirmation of the compound was carried out by identifying using <sup>1</sup>H-NMR.

10 <sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.25-7.09 (m, 2H), 7.03 (m, 1H), 6.90 (m, 1H), 6.49 (m, 1H), 3.78 (s, 3H), 2.55 (s, 3H)

Step 3

15 Production of 1,4-dimethyl-3-(N,N-dimethylaminomethyl)indole

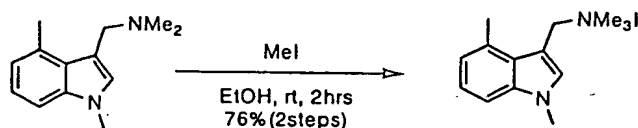


5.9 ml (72.0 mmol) of 37% aqueous formaldehyde solution and 7.08 ml (78 mmol) of 50% aqueous dimethylamine solution were sequentially added to a mixed  
20 system containing 25 ml each of 1,4-dioxane and acetic acid. After cooling to room temperature, as this reaction generates heat, 10 ml of a 1,4-dioxane solution containing 9.24 g (63.6 mmol) of the 1,4-dimethylindole obtained in Step 2 were added followed by stirring for 2  
25 hours at room temperature. The reaction solution was then concentrated as is. 5 M aqueous sodium hydroxide solution were then added to the residue to make alkaline (pH = 12) and bring to a total volume of 100 ml followed by extraction with ethyl acetate (100 ml x 2). The  
30 organic phase was then dried with anhydrous magnesium sulfate and concentrated under reduced pressure to obtain 12.93 g (63.9 mmol) of the target compound (crude yield: 100.4%). Confirmation of the compound was carried out by identifying using <sup>1</sup>H-NMR.

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.15-7.06 (m, 2H), 6.91 (m, 1H), 6.85 (m, 1H), 3.71 (s, 3H), 3.59 (s, 2H), 2.74 (s, 3H), 2.27 (s, 6H)

Step 4

5 Production of ((1,4-dimethylindole-3-yl)methyl)trimethylammonium iodide

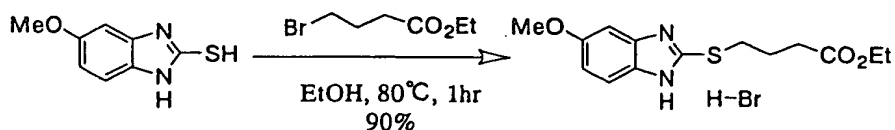


12.93 g (63.6 mmol) of the 1,4-dimethyl-3-(N,N-dimethylaminomethyl)indole obtained in Step 3 were dissolved in 60 ml of ethanol followed by the addition of 4.36 ml (70 mmol) of methyl iodide. A white precipitate formed after stirring for 2 hours at room temperature. This was then filtered, washed twice with 10 ml of ethanol and dried in a vacuum to obtain 16.66 g (48.4 mmol) of the target compound (yield of the two steps: 76%). Confirmation of the compound was carried out by identifying using <sup>1</sup>H-NMR.

<sup>1</sup>H-NMR (270 MHz, DMSO) (ppm): 7.65 (s, 1H), 7.36 (d, 1H), 7.13 (t, 1H), 6.91 (d, 1H), 4.74 (s, 2H), 3.82 (s, 3H), 3.01 (s, 9H), 2.65 (s, 3H)

[Reference Example 7]

20 Production of 4-(5-methoxybenzimidazole-2-ylthio)butanoate ester hydrogen bromide salt

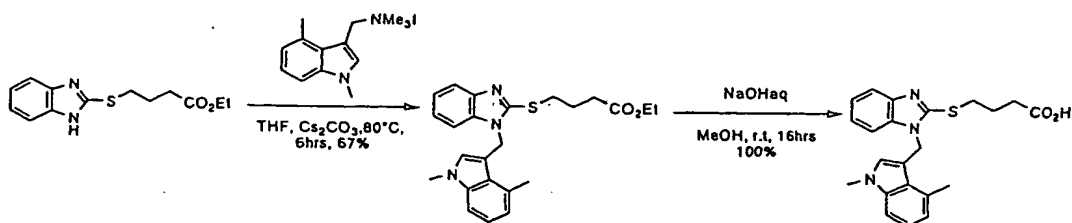


6.48 g (33.2 mmol) of 4-bromobutanoate ethyl ester were added to 10 ml of an ethanol solution containing 5.0 g (27.7 mmol) of 5-methoxybenzimidazole-2-thiol followed by stirring for 1 hour at 80°C and adding 90 ml of ethyl acetate. The reaction solution was returned to room temperature and the formed crystals were filtered out followed by drying to obtain 9.34 g of the target compound (yield: 90%).

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.65 (d, 1H, J=8.91 Hz), 7.24 (s, 1H), 7.00 (dd, 1H, J=2.43, 8.91 Hz), 4.21 (q, 2H, J=7.29 Hz), 3.83 (s, 3H), 3.74 (m, 2H), 2.61 (m, 2H), 2.10 (m, 2H), 1.30 (t, 3H, J=7.29 Hz)

[Example 1]

Production of Compound No. 39



480 mg (2.49 mmol) and 10 ml of tetrahydrofuran were added to a pre-dried reaction vessel. 505 mg (1.91 mmol) of the 4-(benzimidazole-2-ylthio)butanoate ethyl ester obtained in Reference Example 3 and 724 mg (2.10 mmol) of ((1,4-dimethylindole-3-yl)methyl)trimethylammonium iodide were added followed by stirring for 6 hours at 80°C. After filtering the solution by passing through Celite, it was concentrated under reduced pressure. The residue was then purified by silica gel column chromatography (dichloromethane:ethyl acetate = 8:1) to obtain 540 mg (1.28 mmol) of 4-(1-((1,4-dimethylindole-3-yl)methyl)benzimidazole-2-ylthio)butanoate ethyl ester (yield: 67%).

2.0 ml of a 2M aqueous sodium hydroxide solution were then added to 6 ml of a methanol solution containing 540 mg (1.28 mmol) of the resulting 4-(1-((1,4-dimethylindole-3-yl)methyl)benzimidazole-2-ylthio)butanoate ethyl ester. After stirring for 16 hours at room temperature, 6 M hydrochloric acid was added to stop the reaction. The solvent was removed to a certain degree by concentration under reduced pressure followed by extraction with ethyl acetate. After washing the ethyl acetate phase with saturated brine, it was dried with anhydrous magnesium sulfate. After distilling off the solvent under reduced pressure, it was purified

by silica gel column chromatography  
(dichloromethane:methanol = 8:1) to obtain 502 mg (1.28  
mmol) of the target compound (yield: 100%). Confirmation  
of the compound was carried out by identifying from its  
5 molecular weight using LC-MS.

Calculated value  $M = 393.15$ , Measured value  $(M+H)^+ = 394.2$

[Example 2]

The following compounds and the compounds in the  
10 following table were synthesized according to the same  
method as Example 1 using the compounds indicated in  
Reference Example 2 or 3 as well as various quaternary  
ammonium salts or halide derivatives synthesized with  
reference to Reference Examples 4-6 and other references  
15 described in the text. Confirmation of the compounds was  
carried out by identifying from their molecular weights  
using LC-MS. However, some of the compounds were  
synthesized using conditions that somewhat differed from  
those of Example 1, including conditions such as the use  
20 of DMF and so forth for the solvent and the use of  
potassium carbonate for the base in coupling, the use of  
THF and EtOH for the solvent in hydrolysis, and the use  
of a temperature of room temperature to 50°C.

In addition, the following compounds were similarly  
25 synthesized.

4-(1-(2-(1-methylindole-3-yl)ethyl)benzimidazole-2-  
ylthio)butanoic acid (Compound No. 1153)

In this case however, a methanesulfonate ester of 2-  
(1-methylindole-3-yl)ethanol was used instead of  
30 quaternary ammonium salt and halide derivative.  
Identification of the compound was carried out using LC-  
MS. The yield was 19% (two steps of N-alkylation and  
ester hydrolysis).

Calculated value  $M = 393.15$ , Measured value  $(M+H)^+ =$   
35 394.0

4-(1-(4-methyl-7-chlorobenzo[b]thiophene-3-yl)methyl)benzimidazole-2-ylthio)butanoic acid  
(Compound No. 1154)

5 Yield: 15% (two steps of N-alkylation and ester hydrolysis)

Calculated value M = 430.06, Measured value (M+H)<sup>+</sup> = 431.2

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 12.17 (br, 1H), 7.63 (d, 1H, J=7.83 Hz), 7.47-7.40 (m, 2H), 7.26 (d, 1H, J=8.10 Hz), 7.22-7.11 (m, 2H), 6.46 (s, 1H), 5.86 (s, 2H), 3.34 (t, 2H, J=7.29 Hz), 2.84 (s, 3H), 2.34 (t, 2H, J=7.29 Hz), 1.94 (m, 2H)

15 4-(1-(4-methyl-7-bromobenzo[b]thiophene-3-yl)methyl)benzimidazole-2-ylthio)butanoic acid  
(Compound No. 1155)

Yield: 56% (two steps of N-alkylation and ester hydrolysis)

Calculated value M = 474.01, Measured value (M+H)<sup>+</sup> = 477.0

20 <sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 12.18 (br, 1H), 7.63 (d, 1H, J=7.56 Hz), 7.53 (d, 1H, J=7.56 Hz), 7.46 (d, 1H, J=7.56 Hz), 7.22-7.11 (m, 3H), 6.46 (s, 1H), 5.85 (s, 2H), 3.34 (t, 2H, J=7.29 Hz), 2.83 (s, 3H), 2.34 (t, 2H, J=7.29 Hz), 1.97 (m, 2H)

Compound No.	Calculated value M	Measured value (M+H) <sup>+</sup>	Yield (two steps) %
35	393.15	394.2	10
36	393.15	394.2	15
37	393.15	394.1	25
38	393.15	394.1	19
39	393.15	394.2	67
40	407.17	408.2	3
41	413.10	414.3	74
42	397.13	398.3	26
43	409.15	410.1	3
44	455.17	456.2	1
45	517.18	518.1	8
46	413.10	414.1	53
47	397.13	398.1	56
48	409.15	410.3	81
49	404.13	405.2	31
50	409.15	410.1	24
52	382.08	383.2	65
54	416.04	417.3	100
56	396.10	397.3	63
58	396.10	397.1	95
59	416.04	417.1	44
63	410.11	411.3	33
514	408.17	408.3	83
519	421.18	422.2	36
521	441.13	442.3	58
532	410.11	411.3	65
534	444.07	445.3	80
536	424.13	425.3	73
538	424.13	425.2	11
615	461.07	462.0	89
628	450.00	451.0	78

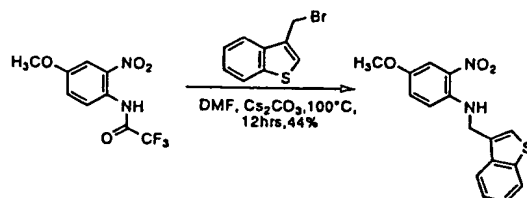
[Example 3]

Production of Compound No. 148

Step 1

5

Production of ((benzothiophene-3-yl)methyl)(4-methoxy-2-nitrophenyl)amine

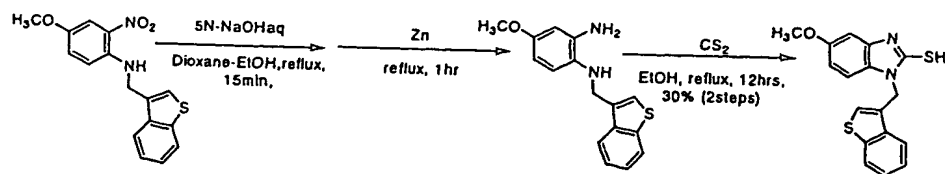


740 mg (2.8 mmol) of 4-methoxy-2-nitrotrifluoroanilide were dissolved in 5 ml of dimethylformamide followed by the sequential addition of

503 mg (3.64 mmol) of potassium carbonate and 773 mg (3.4 mmol) of 3-bromomethylbenzothiophene and heating to 100°C. After 12 hours, 5 ml of 5 M aqueous sodium hydroxide solution were added and refluxed, as is, for 1 hour. After 15 minutes, the solution was cooled to room temperature followed by the addition of 10 ml of water and extraction with chloroform. After washing the organic phase twice with 25 ml of saturated brine and drying with magnesium sulfate, it was concentrated and dried under reduced pressure. The residue was then purified by silica gel column chromatography (hexane:ethyl acetate = 60:1) to obtain 400 mg of ((benzothiophene-3-yl)methyl)(4-methoxy-2-nitrophenyl)amine in the form of an orange powder (yield: 44%).

#### Step 2

#### Production of 1-((benzothiophene-3-yl)methyl)-5-methoxybenzoimidazole-2-thiol



4 ml of ethanol and 4 ml of 1,4-dioxane were added to 400 mg (1.23 mmol) of ((benzothiophene-3-yl)methyl)(4-methoxy-2-nitrophenyl)amine followed by the addition of 0.34 ml of 5 M aqueous sodium hydroxide solution and refluxing while heating. After 15 minutes, the reaction solution was removed from the oil bath followed by the divided addition of 320 mg (4.9 mmol) of zinc powder. The reaction solution was again refluxed while heating for 1 hour. After allowing to cool to room temperature, the zinc was filtered out and the filtrate was concentrated under reduced pressure followed by extraction with chloroform. The organic phase was washed twice with 5 ml of saturated brine followed by drying

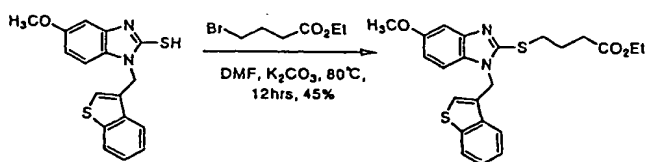


with magnesium sulfate, concentration under reduced pressure and drying to obtain 309 mg of a brown oil.

Continuing, the resulting brown oil was dissolved in 10 ml of ethanol followed by the addition of 2.5 ml (42 mmol) of carbon disulfide and refluxing. After 12 hours, the reaction solution was returned to room temperature and concentrated under reduced pressure followed by the addition of 2 ml of ethanol and irradiating with ultrasonic waves to break into fine fragments that were then filtered. The resulting powder was washed twice with 2 ml of ethanol and then dried to obtain 120 mg (0.37 mmol) of 1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-thiol (yield of the two steps: 30%).

Step 3

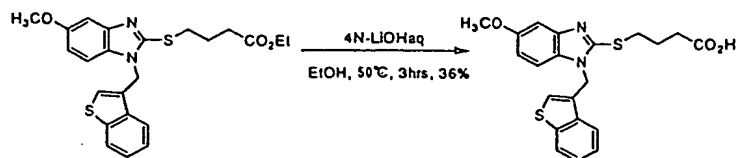
Production of 4-(1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester



101 mg (0.30 mmol) of 1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-thiol were dissolved in 2 ml of dimethylformamide followed by the addition of 62 mg (0.45 mmol) of potassium carbonate and 53 mg (0.40 mmol) of 4-bromobutanoate ethyl ester and heating to 80°C. After 12 hours, the reaction solution was concentrated under reduced pressure and extracted with diethyl ether followed by washing twice with 10 ml of saturated brine and drying with magnesium sulfate. The solvent was then concentrated under reduced pressure and the residue was purified by silica gel column chromatography (hexane:ethyl acetate = 1:1) to obtain 60 mg (0.136 mmol) of 4-(1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester (yield: 45%).

Step 4

Production of 4-(1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoic acid



60 mg (0.136 mmol) of 4-(1-((benzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester were dissolved in 2 ml of methanol followed by the addition of 0.5 ml of 4 M aqueous sodium hydroxide solution. After stirring for 3 hours at 50°C, 6 M hydrochloric acid was added to stop the reaction followed by concentrating under reduced pressure and extracting with chloroform. After washing the organic phase with saturated brine, it was dried with anhydrous magnesium sulfate. The solvent was concentrated under reduced pressure and the residue was purified by silica gel column chromatography (ethyl acetate) to obtain 20 mg (0.048 mmol) of the target compound (yield: 36%). Confirmation of the compound was carried out by identifying from the molecular weight using LC-MS.

Calculated value  $M = 412.09$ , Measured value  $(M+H)^+ = 413.1$

[Example 4]

Production of Compound No. 135

The target compound was obtained according to the same method as Example 3.

However, ((1,4-dimethylindole-3-yl)methyl) trimethylammonium iodide was used in the reaction corresponding to Step 1.

Confirmation of the compound was carried out by identifying from the molecular weight using LC-MS.

Calculated value  $M = 423.16$ , Measured value  $(M+H)^+ = 424.3$

Production of Compound No. 137

The target compound was obtained according to the

same method as Example 3.

However, ((1-methyl-4-chloroindole-3-yl)methyl) trimethylammonium iodide was used in the reaction corresponding to Step 1.

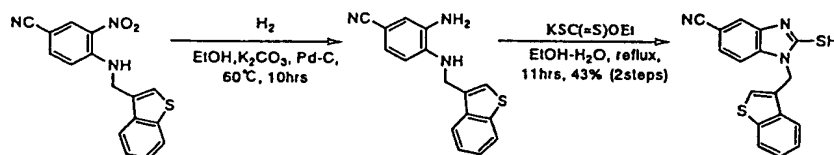
Confirmation of the compound was carried out by identifying from the molecular weight using LC-MS.

Calculated value  $M = 443.11$ , Measured value  $(M+H)^+ = 444.3$

[Example 5]

Production of Compound No. 244

The target compound was obtained using the same method as Example 3. However, 4-cyano-2-nitrotrifluoroacetonitrile was used as the reagent corresponding to Step 1. In addition, the step in which the 2-nitroaniline derivative is reduced to an orthophenylenediamine derivative, and the step in which this is cyclized to a benzimidazole-2-thiol derivative were carried out using the methods described below.



10 ml of ethanol were added to 1.1 g (3.56 mmol) of ((3-benzothiophenyl)methyl)(4-cyano-2-nitrophenyl)amine followed by the addition of 2.4 g (17.8 mmol) of potassium carbonate. After replacing the reaction system with nitrogen, 220 mg of 10% palladium-carbon were added followed by replacing the reaction system with hydrogen and heating to  $60^\circ C$ .

After 4 hours and 30 minutes, an additional 220 mg of 10% palladium-carbon were added followed by replacing the reaction system with hydrogen and heating to  $60^\circ C$ . 5 hours and 10 minutes after the start of the reaction, the reaction system was cooled to room temperature. The reaction solution was then filtered with Celite and concentrated under reduced pressure to obtain 0.93 g of a

liquid residue. Continuing, 0.93 g (2.63 mmol) of ((2-benzothiophenyl)methyl)(2-amino-4-methylphenyl)amine were dissolved in 10 ml of ethanol and 2 ml of water followed by refluxing after adding 2.1 g (13.3 mmol) of potassium ethylxanthate. After 11 hours, 12.5 ml of 40% aqueous acetic acid solution were dropped in. After cooling to room temperature and concentrating under reduced pressure, the residue was purified by silica gel column chromatography (hexane:acetone = 2:1) to obtain 491.7 mg of 1-((2-benzothiophenyl)methyl)-6-cyanobenzimidazole-2-thiol (yield of the two steps: 43%). Confirmation of compound no. 244 was carried out by identifying from the molecular weight using <sup>1</sup>H-NMR and LC-MS.

Calculated value M = 407.08, Measured value (M+H)<sup>+</sup> = 408.2

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) (ppm): 7.94 (s,1H), 7.76 (dd,1H), 7.52 (dd,1H), 7.42 (m,3H), 7.31 (d,1H), 7.00 (s,1H), 5.56 (s,2H), 3.35 (t,2H), 2.47 (t,2H), 2.15 (p,2H)

[Example 6]

The following target compounds were obtained using the same method as Example 5.

Production of Compound No. 340

4-methyl-2-nitrotrifluoroacetoanilide was used as the reagent corresponding to Step 1.

Confirmation of compound no. 340 was carried out by identifying from the molecular weight using LC-MS.

Calculated value M = 396.10, Measured value (M+H)<sup>+</sup> = 397.0

Production of Compound No. 436

5-methyl-2-nitrotrifluoroacetoanilide was used as the reagent corresponding to Step 1.

Confirmation of compound no. 436 was carried out by identifying from the molecular weight using LC-MS.

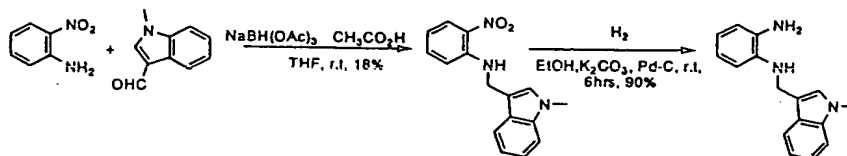
Calculated value M = 396.10, Measured value (M+H)<sup>+</sup> = 397.0

[Example 7]

Production of Compound No. 34

Step 1

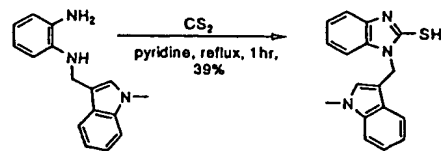
Production of ((1-methylindole-3-yl)methyl)(2-aminophenyl)amine



5            829 mg (6 mmol) of 2-nitroaniline and 1242 mg (7.8 mmol) of 1-methylindole carboxyaldehyde were dissolved in 20 ml of tetrahydrofuran followed by the sequential addition of 200  $\mu$ l of acetic acid and 5087 mg (24 mmol) of NaBH(OAc)<sub>3</sub> and stirring overnight at room temperature. 10 After adding saturated aqueous sodium bicarbonate solution, extracting with ethyl acetate and drying with anhydrous magnesium sulfate, the solvent was distilled off and the residue was purified by silica gel column chromatography (hexane:ethyl acetate = 95:5) to obtain 15 264 mg of ((1-methylindole-3-yl)methyl)(2-nitrophenyl)amine (yield: 18%). 264 mg (0.939 mmol) of ((1-methylindole-3-yl)methyl)(2-nitrophenyl)amine were then dissolved in 10 ml of ethanol followed by the addition of 50 mg (0.047 mmol) of 10% Pd-C and stirring 20 for 6 hours at room temperature in a hydrogen atmosphere. After completion of the reaction, the Pd-C was filtered out and the solvent was distilled off under reduced pressure to obtain 212 mg of ((1-methylindole-3-yl)methyl)(2-aminophenyl)amine (yield: 90%).

25            Step 2

Production of 1-((1-methylindole-3-yl)methyl)benzimidazole-2-thiol

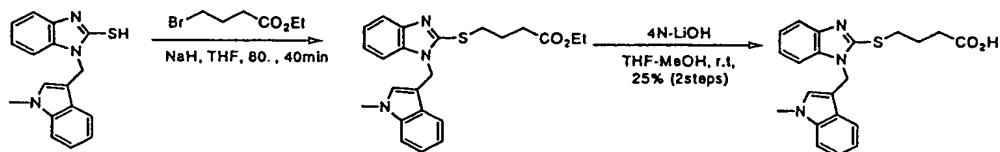


212 mg (0.845 mmol) of ((1-methylindole-3-

yl)methyl)(2-aminophenyl)amine were dissolved in 1 ml of pyridine followed by the addition of 1 ml (16.9 mmol) of carbon disulfide and refluxing for 1 hour in a nitrogen atmosphere. The solvent was distilled off followed by purification by silica gel column chromatography (hexane:ethyl acetate = 2:1) to obtain 96 mg of 1-((1-methylindole-3-yl)methyl)benzimidazole-2-thiol (yield: 39%).

Step 3

Production of 4-(1-((1-methylindole-3-yl)methyl)benzimidazole-2-ylthio)butanoic acid



12 mg (0.342 mmol) of sodium hydride and 2 ml of tetrahydrofuran were added to a pre-dried reaction vessel. 50 mg (0.171 mmol) of 1-((1-methylindole-3-yl)methyl)benzimidazole-2-thiol and 34  $\mu$ l (0.23 mmol) of 4-bromobutanoate ethyl ester were then added to the reaction vessel followed by stirring for 40 minutes at 60°C. Water was then added followed by extraction with ethyl acetate. After drying the ethyl acetate phase with anhydrous magnesium sulfate, the reaction solution was concentrated under reduced pressure and the resulting residue was purified by silica gel column chromatography (hexane:ethyl acetate = 3:1) to obtain 4-(1-((1-methylindole-3-yl)methyl)(benzimidazole-2-ylthio)butanoate ethyl ester. Continuing, 0.25 ml of 4 M aqueous lithium hydroxide solution were added to 1 ml of tetrahydrofuran containing this 4-(1-((1-methylindole-3-yl)methyl)(benzimidazole-2-ylthio)butanoate ethyl ester and 0.5 ml of methanol. After stirring overnight at room temperature, 6 M hydrochloric acid was added to stop the reaction followed by extraction with ethyl acetate. After washing the ethyl acetate phase with saturated brine, it was dried with anhydrous magnesium sulfate.

The solvent was then distilled off under reduced pressure to obtain 16 mg (0.042 mmol) of the target compound (yield: 25%).

Confirmation of the compound was carried out by identifying from the molecular weight using LC-MS.

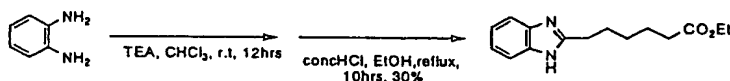
Calculated value  $M = 379.14$ , Measured value  $(M+H)^+ = 380.2$

[Example 8]

Production of Compound No. 743

Step 1

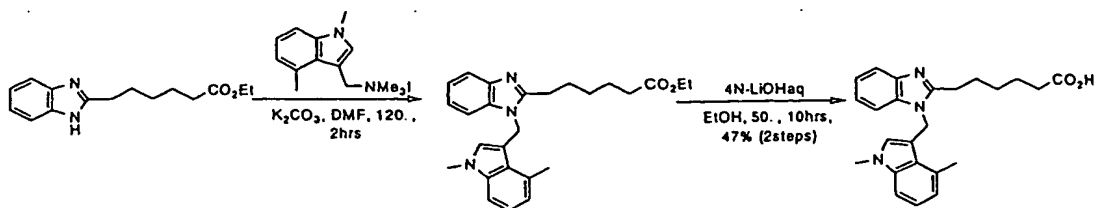
Production of 5-(benzimidazole-2-yl)pentanoate ethyl ester



696  $\mu$ l (5.0 mmol) of triethylamine and 893 mg (5.0 mmol) of methyladipochloride were dropped into 10 ml of a chloroform solution containing 540 mg (5.0 mmol) of orthophenylenediamine followed by stirring for 12 hours at room temperature. 20 ml of ethanol and 4 ml of concentrated hydrochloric acid were then added followed by stirring for 10 hours while heating and refluxing. The reaction solution was then neutralized using 5 M aqueous sodium hydroxide solution followed by extraction with ethyl acetate. After washing with water and concentrating under reduced pressure, the residue was purified by silica gel column chromatography (ethyl acetate only) to obtain 359 mg of 5-(benzimidazole-2-yl)pentanoate ethyl ester (yield: 30%).

Step 2

Production of 5-(1-((1,4-dimethylindole-3-yl)methyl)benzimidazole-2-yl)pentanoic acid

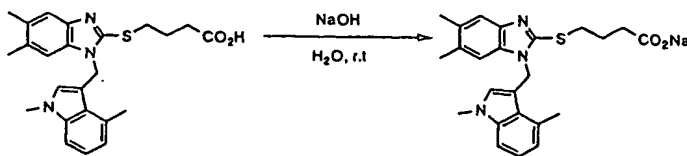


42 mg (0.3 mmol) of potassium carbonate and 103 mg (0.3 mmol) of ((1,4-dimethylindole-3-yl)methyl)trimethylammonium iodide were added to 2 ml of DMF solution containing 50 mg (0.2 mmol) of the resulting 5-(benzimidazole-2-yl)pentanoate ethyl ester followed by stirring for 2 hours at 120°C. The resulting solution was extracted with dichloromethane, washed with water and concentrated followed by purification of the residue by column chromatography (hexane:ethyl acetate = 1:2). 5 ml of ethanol and 0.5 ml of 4 M aqueous sodium hydroxide solution were then added to this followed by stirring for 10 hours at 50°C and then the addition of 6 M hydrochloric acid to stop the reaction. The solution was extracted with chloroform, and after washing with water and concentrating under reduced pressure, the residue was purified by silica gel column chromatography (chloroform:methanol = 10:1) to obtain 35 mg of the target compound (yield of the two steps: 47%). Confirmation of the compound was carried out by identifying from the molecular weight using LC-MS.

Calculated value  $M = 375.19$ , Measured value  $(M+H)^+ = 376.5$

#### [Example 9]

#### Production of Sodium Salt of Compound No. 519



11.9 ml (1.19 mmol) of 0.1 M aqueous sodium hydroxide solution were added to 100 ml of an aqueous solution containing 503 mg (1.19 mmol) of the above compound no. 519 followed by stirring at room



temperature. Subsequently, the reaction solution was freeze-dried to obtain 470 mg (1.05 mmol) of the sodium salt (yield: 89%).

5  $^1\text{H}$ -NMR (400 MHz, DMSO- $d_6$ ) (ppm): 7.37 (s,1H), 7.19 (d,1H, $J$ =8.24 Hz), 7.09-7.01 (m,2H), 6.80 (d,1H, $J$ =7.09 Hz), 6.32 (s,1H), 5.66 (s,2H), 3.59 (s,3H), 3.26 (m,2H), 2.66 (s,3H), 2.27 (s,3H), 2.21 (s,3H), 1.95 (m,2H), 1.81 (m,2H)

[Example 10]

10 The compounds indicated below were synthesized using the respective corresponding substrates according to the same method as Example 9.

Sodium Salt of Compound No. 39

15  $^1\text{H}$ -NMR (270 MHz, DMSO- $d_6$ ) (ppm): 7.57 (d,1H, $J$ = Hz), 7.28 (d,1H, $J$ =7 Hz), 7.20 (d,1H, $J$ =8 Hz), 7.15-7.00 (m,3H), 6.77 (d,1H, $J$ =7 Hz), 6.47 (s,1H), 5.69 (s,2H), 3.60 (s,3H), 3.31 (t,2H, $J$ =7 Hz), 2.61 (s,3H), 1.99 (t,2H, $J$ =7 Hz), 1.84 (p,2H, $J$ =7 Hz)

Sodium Salt of Compound No. 52

20  $^1\text{H}$ -NMR (400 MHz, DMSO- $d_6$ ) (ppm): 7.97 (d,1H), 7.91 (d,1H, $J$ =6.76 Hz), 7.57 (d,1H, $J$ =7.75 Hz), 7.44-7.38 (m,3H), 7.30 (s,1H), 7.12 (m,2H), 5.63 (s,2H), 3.33 (m,2H), 2.03 (m,2H), 1.87 (m,2H)

Sodium Salt of Compound No. 135

25  $^1\text{H}$ -NMR (400 MHz, DMSO- $d_6$ ) (ppm): 7.21-7.00 (m,4H), 6.79 (d,1H, $J$ =7.29 Hz), 6.67 (dd,1H, $J$ =2.43, 8.91 Hz), 6.51 (s,1H), 5.65 (s,2H), 3.75 (s,3H), 3.62 (s,3H), 3.31 (m,2H), 2.59 (s,3H), 1.95 (m,2H), 1.82 (m,2H)

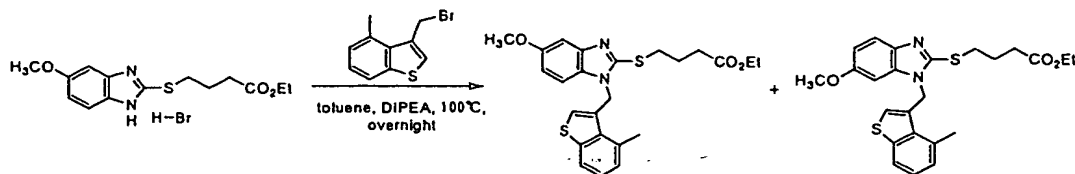
Sodium Salt of Compound No. 532

30  $^1\text{H}$ -NMR (400 MHz, DMSO- $d_6$ ) (ppm): 7.98 (d,1H, $J$ =7.42 Hz), 7.90 (d,1H, $J$ =6.43 Hz), 7.44-7.39 (m,2H), 7.35 (s,1H), 7.18 (m,2H), 5.57 (s,2H), 3.28 (m,2H), 2.26 (s,3H), 2.23 (s,3H), 1.99 (m,2H), 1.84 (m,2H)

[Example 10]

Production of 4-(1-((4-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester and 4-(1-((4-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoate ethyl ester

5



539 mg (1.44 mmol) of 4-(5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester were suspended in 4 ml of toluene followed by the addition of 616  $\mu$ l (3.60 mmol) of diisopropylethylamine and 384 mg (1.59 mmol) of 4-methyl-3-(bromomethyl)benzo[b]thiophene and heating at 100°C. After allowing to react overnight, saturated sodium bicarbonate solution was added followed by extraction with ethyl acetate. The organic phase was washed with water followed by drying with magnesium sulfate and concentrating the solvent under reduced pressure. The resulting residue was purified by silica gel column chromatography (hexane:ethyl acetate = 4:1) to obtain 114 mg of 4-(1-((4-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester (yield: 17%) and 68 mg of 4-(1-((4-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoate ethyl ester (yield: 10%).

4-(1-((4-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester

25

$^1\text{H-NMR}$  (270 MHz,  $\text{CDCl}_3$ ) (ppm): 7.71 (d, 1H,  $J=7.56$  Hz), 7.62 (d, 1H,  $J=8.64$  Hz), 7.30-7.18 (m, 2H), 6.87 (dd, 1H,  $J=2.43$ , 8.64 Hz), 6.61 (d, 1H,  $J=2.43$  Hz), 6.42 (s, 1H), 5.74 (s, 2H), 4.10 (q, 2H,  $J=7.29$  Hz), 3.75 (s, 3H), 3.38 (t, 2H,  $J=7.29$  Hz), 2.89 (s, 3H), 2.45 (t, 2H,  $J=7.29$  Hz), 2.11 (m, 2H), 1.23 (t, 3H,  $J=7.29$  Hz)

30

4-(1-((4-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoate ethyl ester

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.70 (d, 1H, J=8.10 Hz), 7.29-7.17 (m, 3H), 7.02 (d, 1H, J=8.91 Hz), 6.80 (dd, 1H, J=2.43, 8.91 Hz), 6.40 (s, 1H), 5.74 (s, 2H), 4.11 (q, 2H, J=7.29 Hz), 3.87 (s, 3H), 3.42 (t, 2H, J=7.02 Hz), 2.88 (s, 3H), 2.46 (t, 2H, J=7.29 Hz), 2.10 (m, 2H), 1.23 (t, 3H, J=7.29 Hz)

[Example 11]

The following compounds were obtained according to the same method as Example 10.

4-(1-((5-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester  
(Yield: 24%)

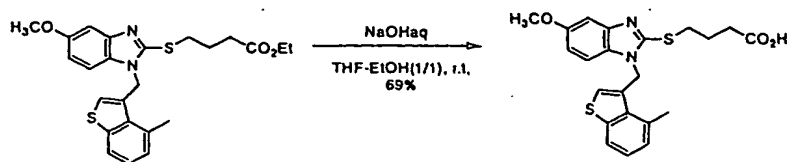
<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.76 (d, 1H, J=8.10 Hz), 7.62 (s, 1H), 7.58 (d, 1H, J=8.64 Hz), 7.25 (1H), 6.84 (dd, 1H, J=2.43, 8.91 Hz), 6.81 (s, 1H), 6.65 (d, 1H, J=2.16 Hz), 5.47 (s, 2H), 4.11 (q, 2H, J=7.02 Hz), 3.74 (s, 3H), 3.39 (t, 2H, J=7.02 Hz), 2.51 (s, 3H), 2.47 (t, 2H, J=7.56 Hz), 2.11 (m, 2H), 1.24 (t, 3H, J=7.02 Hz)

4-(1-((5-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoate ethyl ester  
(Yield: 18%)

<sup>1</sup>H-NMR (270 MHz, CDCl<sub>3</sub>) (ppm): 7.75 (d, 1H, J=8.10 Hz), 7.60 (s, 1H), 7.26-7.22 (m, 2H), 7.04 (d, 1H, J=8.91 Hz), 6.83 (s, 1H), 6.78 (dd, 1H, J=2.43, 8.91 Hz), 5.47 (s, 2H), 4.12 (q, 2H, J=7.02 Hz), 3.84 (s, 3H), 3.43 (t, 2H, J=7.29 Hz), 2.50 (s, 3H), 2.48 (t, 2H, J=7.29 Hz), 2.12 (m, 2H), 1.24 (t, 3H, J=7.02 Hz)

[Example 12]

Production of 4-(1-((4-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoic acid (Compound No. 154)



84.7 mg (0.186 mmol) of the 4-(1-((4-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoate ethyl ester obtained in Example 10 were dissolved in a mixed solvent of 1 ml of THF and 1 ml of ethanol followed by the addition of 1 ml of 1 M aqueous sodium hydroxide solution and stirring for 1 hour at 40°C. Following completion of the reaction, 1.5 ml of 1 M hydrochloric acid were added followed by stirring for 30 minutes at room temperature. The resulting precipitate was filtered, washed with water, washed with ethanol and then dried to obtain 54.9 mg of the target compound (yield: 69%).

LC-MS:

Calculated value M = 426.11, Measured value (M+H)<sup>+</sup> = 427.2

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 7.80 (d, 1H, J=7.29 Hz), 7.60 (d, 1H, J=8.91 Hz), 7.31-7.20 (m, 3H), 6.95 (dd, 1H, J=2.16, 8.91 Hz), 6.53 (s, 1H), 5.94 (s, 2H), 3.73 (s, 3H), 3.37 (t, 2H, J=7.29 Hz), 2.86 (s, 3H), 2.34 (t, 2H, J=7.29 Hz), 1.90 (m, 2H)

[Example 13]

The following compounds were synthesized according to the same method as Example 12.

4-(1-((4-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoic acid  
(Compound No. 1114)

Yield: 60%

LC-MS:

Calculated value M = 426.11, Measured value (M+H)<sup>+</sup> = 427.2

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 7.78 (d, 1H, J=7.83 Hz), 7.52 (d, 1H, J=8.91 Hz), 7.34-7.17 (m, 3H), 6.77 (dd, 1H, J=2.34, 8.91 Hz), 6.37 (s, 1H), 5.83 (s, 2H), 3.78

(s,3H), 3.32 (t,2H,J=7.29 Hz), 2.82 (s,3H), 2.34 (t,2H,J=7.56 Hz), 1.93 (m,2H)

In this case however, 1 M hydrochloric acid was added following completion of the reaction followed by extraction with chloroform and washing with water. Drying was then performed with magnesium sulfate followed by concentrating the solvent under reduced pressure and drying to obtain the target compound.

4-(1-((5-methylbenzothiophene-3-yl)methyl)-5-methoxybenzimidazole-2-ylthio)butanoic acid  
(Compound No. 152)

Yield: 63%

LC-MS:

Calculated value M = 426.11, Measured value (M+H)<sup>+</sup> = 426.8

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 7.88 (d,1H,J=8.64 Hz), 7.76 (s,1H), 7.58 (d,1H,J=8.64 Hz), 7.28-7.24 (m,3H), 6.94 (dd,1H,J=2.16, 8.64 Hz), 5.72 (s,2H), 3.74 (s,3H), 3.40 (t,2H,J=7.29 Hz), 2.42 (s,3H), 2.36 (t,2H,J=7.29 Hz), 1.92 (m,2H)

4-(1-((5-methylbenzothiophene-3-yl)methyl)-6-methoxybenzimidazole-2-ylthio)butanoic acid  
(Compound No. 1112)

Yield: 79%

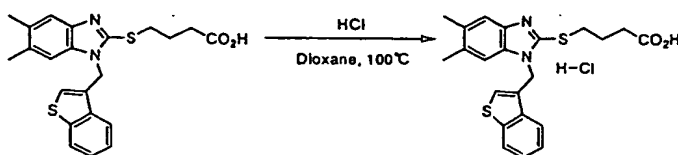
LC-MS:

Calculated value M = 426.11, Measured value (M+H)<sup>+</sup> = 427.0

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 7.87 (d,1H,J=8.10 Hz), 7.71 (s,1H), 7.47 (d,1H,J=8.91 Hz), 7.24 (m,2H), 7.17 (d,1H,J=2.16 Hz), 6.84 (dd,1H), 5.64 (s,2H), 3.77 (s,3H), 3.38 (t,2H,J=7.02 Hz), 2.41 (s,3H), 2.37 (t,2H,J=7.56 Hz), 1.95 (m,2H)

[Example 14]

Production of HCl Salt of Compound No. 532



1.5 ml of 4 M hydrochloric acid/dioxane solution were added to 50 mg (0.122 mmol) of compound no. 532 followed by stirring at 100°C. Following completion of the reaction, the reaction solution was concentrated under reduced pressure to obtain 53 mg (1.05 mmol) of the target compound (yield: 97%).

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 8.00 (m, 1H), 7.89 (m, 1H), 7.52 (m, 2H), 7.45-7.42 (m, 2H), 7.32 (s, 1H), 5.78 (s, 2H), 3.48 (t, 2H, J=7.42 Hz), 2.37 (m, 2H), 2.34 (s, 3H), 2.30 (s, 3H), 1.92 (t, 2H, J=7.09 Hz)

[Example 15]

#### Production of HCl Salt of Compound No. 56

The target compound was obtained according to the same method as Example 14.

<sup>1</sup>H-NMR (270 MHz, DMSO-d<sub>6</sub>) (ppm): 7.87 (d, 1H, J=8.08 Hz), 7.74 (s, 1H), 7.66 (d, 1H, J=6.76 Hz), 7.58 (d, 1H, J=8.74 Hz), 7.26 (m, 4H), 5.70 (s, 2H), 3.45 (t, 2H, J=7.26 Hz), 2.42 (s, 3H), 2.39 (t, 2H, J=7.26 Hz), 1.98 (m, 2H)

[Example 16]

#### Preparation of Recombinant Human Mast Cell Chymase

Recombinant human mast cell chymase was prepared in accordance with the report of Urata, et al. (Journal of Biological Chemistry, Vol. 266, p. 17173 (1991)).

Namely, human mast cell chymase was purified by heparin sepharose (Pharmacia) from a culture supernatant of insect cells (Th5) infected with recombinant baculovirus containing cDNA coding for human mast cell chymase. Moreover, after activating in accordance with the report of Murakami, et al. (Journal of Biological Chemistry, Vol. 270, p. 2218 (1995)), the human mast cell chymase was purified with heparin sepharose to obtain active human mast cell chymase.

[Example 17]

Measurement of Inhibition of Enzyme Activity of  
Recombinant Human Mast Cell Chymase

After adding 2  $\mu$ l of DMSO solution containing a  
5 compound of the present invention to 50  $\mu$ l of Buffer A  
(0.5-3.0 M NaCl, 50 mM Tris-HCl, pH 8.0) containing 1-5  
ng of the active human mast cell chymase obtained in  
Example 16, 50  $\mu$ l of Buffer A containing 0.5 mM succinyl-  
10 alanyl-histidyl-prolyl-phenylalanylparanitroanilide  
(Bacchem) as substrate were added and allowed to react  
for 5 minutes at room temperature. The changes over time  
in absorbance at 405 nm were measured to investigate  
inhibitory activity.

As a result, compound nos. 39, 56, 58, 59, 63, 148,  
15 154, 519, 532, 534, 536, 538, 615, 1112 and 1114 were  
observed to demonstrate inhibitory activity of  $IC_{50}$  = 1  
nM to less than 10 nM, while compound nos. 34, 38, 41,  
42, 52, 54, 135, 137, 152, 244, 340, 436, 514, 521 and  
628 were observed to demonstrate inhibitory activity of  
20  $IC_{50}$  = 10 nM to 100 nM.

As has been shown above, the benzimidazole  
derivatives of the present invention exhibit potent  
chymase inhibitory activity. Thus, the benzimidazole  
derivatives of the present invention were clearly  
25 demonstrated to be human chymase activity inhibitors that  
can be applied clinically for use in the prevention  
and/or treatment of various diseases involving human  
chymase.

[Example 18]

30 Production of Tablets

Tablets were produced having the individual tablet  
composition shown below.

Compound No. 39	50 mg
Lactose	230 mg
35 Potato starch	80 mg
Polyvinylpyrrolidone	11 mg

Magnesium stearate

5 mg

The compound of the present invention (compound of the examples), lactose and potato starch were mixed followed by uniformly wetting with a 20% ethanol solution of polyvinylpyrrolidone, passing through a 20 mesh sieve, drying at 45°C and again passing through a 15 mesh sieve. The granules obtained in this manner were then mixed with magnesium stearate and compressed into tablets.

[Example 19]

Measurement of Blood Concentration During Administration by Intragastric Forced Feeding to Rats

The compounds indicated with the above compound nos. 39, 52 and 244 were administered by intragastric forced feeding to male SD rats while fasting at a dose of 30 mg/kg, after which blood samples were collected immediately after administration and at 30 minutes and 1, 2 and 4 hours after administration. Following collection of blood samples, where samples were immediately separated into serum components, the compound of the present invention was extracted by ordinary solid phase extraction methods, and the resulting samples were analyzed by HPLC using an ODS column (32% acetonitrile-water-0.05% TFA was used for the mobile phase for compound nos. 52 and 244, while 47% acetonitrile-water-10 mM ammonium acetate buffer (pH 4.0) was used for the mobile phase for compound no. 39) followed by measurement of the amount of the unchanged form. Those results are shown in the table below.

Compound No.	After 30 min. (µg/ml)	After 4 hr. (µg/ml)
52	60.5	12.7
244	16.5	8.9
39	16.1	6.3

On the basis of the above results, the compounds of the present invention were rapidly absorbed after administration, and blood concentrations of the unchanged form shown in the table were measured after 30 minutes.



Moreover, although blood concentrations decreased gradually until 4 hours after administration, a considerable amount of the unchanged forms could still be confirmed even at 4 hours after administration. Thus, the compounds of the present invention were determined to be a group of compounds having superior pharmacokinetic properties. The pharmacokinetic properties of the group of compounds in which A is  $-\text{CH}_2\text{CH}_2\text{CH}_2-$  are particularly superior.

[Example 20]

In Vitro Metabolism Test Using Liver Microsomes (Ms)

Measurement Method:

\* Reaction Solution Composition and Reaction Conditions

Reconstruction system Composition	Composition	Composition and Procedure		Comments
	Buffer	Reagent name	Final conc.	Reaction solution volume: 0.5 mL
		Phosphate buffer (pH 7.4)	0.1 M	
	Chelating agent	EDTA	1.0 mM	
	NADPH generation system	Magnesium chloride	3.0 mM	
		G6P	5.0 mM	
		G6PDH	1.0 IU	
	Enzyme	Liver microsomes	1.0 mg/mL	
	Substrate	Substrate (evaluation compound)	5.0 $\mu\text{M}$	
	Reaction initiator	NADPH	1.0 mM	
Reaction conditions		37°C, incubation (water bath, shaking), reaction times: 0, 2, 5, 10 and 30 min.		
Reaction terminator (extraction liquid)		Acetonitrile		Equal to 3 volumes of reaction solution
Deproteinization		Sampling of supernatant after centrifuging for 10 min. at 3000 rpm, removal of solvent with evaporator		
Redissolution liquid		Redissolution with HPLC mobile phase used for analysis		
Analysis		Detection of peak of unchanged form by HPLC using UV detector		

\* MR Calculation Method

The metabolic rate was determined from the decrease in the amount of the unchanged form at each reaction time and the reaction time based on assigning a value of 100% to the amount of the unchanged form at the initial concentration (reaction time: 0 minutes), and the metabolic rate at the time the metabolic rate reached a maximum was evaluated as the MR value.

MR = (substrate concentration at reaction time: 0 min. - substrate concentration after reaction) ÷ reaction time ÷ protein concentration (nmol/min./mg protein)

These methods were used to obtain the measurement results indicated below.

Compound No.	MR	Percentage of substrate remaining after 30 min. (%)
34	0.260	60.3
38	0.329	29.8
39	0	80.1
41	0.129	73.9
52	0.331	47.5
56	0.111	41.2
58	0.048	72.3
135	0.097	55.2
244	0.211	57.9
514	0.087	48.7
519	0.102	52.9
521	0.088	61.1
532	0.277	36.2
534	0.102	63.0
536	0.131	56.3
615	0.159	62.3

According to the above results, the compounds of the present invention are a group of metabolically stable compounds. The group of compounds in which A is -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>- was determined to be a group of particularly metabolically stable compounds.

Industrial Applicability

The benzimidazole derivatives of the present invention or their medically allowed salts exhibit potent human chymase inhibitory activity. Thus, said

benzimidazole derivatives or their medically allowed salts can be used as preventive and/or therapeutic agents that can be applied clinically as human chymase inhibitors for inflammatory diseases, allergic diseases, respiratory diseases, cardiovascular diseases or bone and cartilage metabolic diseases.